

AQA Chemistry

GCSE Exam-style mark scheme

C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark	Mark	Mark
10.1.1	14.2 g		1	100
10.1.2	14.2 g	Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$	10	100
10.1.3	any 1000 factor - converting grams - into kg (multiplication) - needed - into kg (multiplication) needed - into the equation		10	100
10.2	all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct	Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$	10	100
10.3.1	all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct	Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$	10	100
10.3.2	all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct	Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$	10	100
10.3.3	all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct	Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$	10	100

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JS Bruner



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