

Science 101

Chapter Test A

Teacher Notes and Answers

Subatomic Physics

Questions 1-10 (Answers)

1. a
 2. b
 3. c
 4. d
 5. e
 6. f
 7. g
 8. h
 9. i
 10. j
- Answers:
1. Mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
2. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
3. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
4. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
5. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
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8. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
9. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
10. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.

11. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
12. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
13. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
14. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
15. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
16. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
17. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
18. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
19. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
20. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.

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J Sowinski, S E Vigdor



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