

5. Analyze: What pattern, if any, do you see in your data?

I hardly notice any differences save the general decline.

6. Revise and repeat: Use your data from #4 above to fill in the first line of the data table below. Then repeat the experiment four more times. Calculate the average number of radioactive atoms for each time.

| Trial     | 0 s | 10 s | 20 s | 30 s | 40 s | 50 s |
|-----------|-----|------|------|------|------|------|
| 1         | 128 | 55   | 28   | 11   | 6    | 3    |
| 2         | 128 | 71   | 33   | 17   | 11   | 7    |
| 3         | 128 | 58   | 27   | 16   | 4    | 3    |
| 4         | 128 | 64   | 32   | 17   | 9    | 5    |
| 5         | 128 | 60   | 31   | 12   | 5    | 2    |
| Averages: | 128 | 62   | 30   | 15   | 7    | 4    |

7. Analyze: A **half-life** is defined as the amount of time it takes for half of the radioactive particles to decay. For the simulated substance, every 10 seconds represents one half-life.

How does your data demonstrate the definition of a half-life?

On average, it is half of the previous one.

8. Revise and repeat: Click **Reset**. Real radioactive samples will contain billions of radioactive atoms. To model the decay of a large sample, change from **Random decay** to **Theoretical decay** on the SIMULATION pane. Click **Play** and record the numbers of radioactive atoms:

0 s: 128    10 s: 64    20 s: 32    30 s: 16    40 s: 8    50 s: 4

How does this data demonstrate the meaning of half-life?

Every life is exactly half of the previous one.

# Half Life Gizmo Answer Key

**M Walker**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'M Walker' text.

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