

HONORS PHYSICAL SCIENCE

FINAL EXAM STUDY GUIDE

- Science Skills
 - Explain how science and technology are related
 - Describe the steps of the scientific method
 - Compare and contrast accuracy and precision
 - Perform calculations involving scientific notation and both temperature and metric conversion factors
 - Identify the relationship between a manipulated (independent) variable and a responding (dependent) variable
 - Compare and contrast a scientific law and a scientific theory
- Properties and States of Matter
 - Compare and contrast the characteristics of elements, compounds, pure substances, mixtures (heterogeneous and homogeneous), solutions, suspensions, and colloids
 - Understand the methods of separation
 - Compare and contrast physical and chemical properties, and physical and chemical changes
 - Explain the behavior of gases, liquids, and solids using kinetic theory
 - Predict changes in gas pressure due to changes in temperature, volume, and number of particles
 - Explain Charles's Law, Boyle's Law, and the combined gas law
 - Apply gas laws to solve problems involving gases
 - Name and describe each of the six phase changes. Which are endothermic and which are exothermic?
- Atoms, Elements, and Bonds
 - Compare and contrast Bohr and Rutherford's atomic models
 - Identify three subatomic particles and compare their properties
 - Compare and contrast atoms in ground and excited states
 - Distinguish the atomic number of an element from the mass number of an isotope, and use these numbers to describe the structure of atoms
 - Explain how the electron cloud model represents the behavior and locations of electrons in atoms
 - Understand the development and organization of the periodic table, especially in terms of group properties, atomic mass and valence electrons
 - Identify general properties of metals, nonmetals, and metalloids
 - Predict the reactivity of some elements based on their locations in the periodic table
 - Compare and contrast ionic and covalent bonds
 - Understand the concept of polarity and the role it plays in bonding
- Chemical Reactions, Acids, and Bases
 - Interpret chemical equations in terms of reactants, products, and conservation of mass
 - Balance chemical equations by manipulating coefficients
 - Convert between moles and mass of a substance using molar mass
 - Calculate amounts of reactants or products by using molar mass, mole ratios, and balanced chemical equations
 - Classify chemical reactions as synthesis, decomposition, single replacement, double replacement, or combustion reactions
 - Explain how energy is conserved during chemical reactions
 - Describe the factors affecting chemical reaction rates
 - Describe the factors affecting chemical equilibrium (see Chapter's Principles)
 - Define acid and describe some of the general properties of an acid
 - Define base and define some of the general properties of a base
 - Understand what a buffer is and how it can be used
 - Know what a neutralization reaction is, and what the products are
- Nuclear Chemistry
 - Describe the process of nuclear decay

Honors Physical Science Final Exam Study Guide

Anthony S. Fauci



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Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Honors Physical Science Final Exam Study Guide**

In a global inundated with screens and the cacophony of instant connection, the profound power and emotional resonance of verbal art usually fade in to obscurity, eclipsed by the regular barrage of sound and distractions. Yet, located within the musical pages of **Honors Physical Science Final Exam Study Guide**, a charming function of literary beauty that pulses with raw feelings, lies an unique journey waiting to be embarked upon. Penned by a virtuoso wordsmith, this mesmerizing opus manuals readers on a psychological odyssey, softly exposing the latent potential and profound affect embedded within the intricate internet of language. Within the heart-wrenching expanse of this evocative evaluation, we can embark upon an introspective exploration of the book is key subjects, dissect their captivating writing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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