Honors Chemistry Worksheet 3 Stoichiometry Practice Problems

Name	Period	Date

Instructions: Balance the following chemical equations and then determine the missing information for each of the conditions given. The four questions related to each equation are independent of one another. Answers for a particular numbered problem should be in the same units as the information given (i.e. grams to grams, moles to moles, particles to particles).

	$AJ_2O_{3(s)}$	-	$AI_{(n)}$	+:		$O_{2(g)}$	Work	Area
1.	5mol_							
2.	(*) 		2.5mol					
3.	25g_					<u></u> -		
4.	-					2.41x10 ²⁴ molecules		
	CuO ₍₁₎		H ₂ SO _{4(aq)}		\rightarrow	CuSO _{4(aq)}	+	$H_2O_{(1)}$
1.			.75mol					
2.	50g							
3.						3.5mol		
4.								1.204×10 ³³ molecules
	K ₍₁₀₎		$N_{2(g)}$		>	$K_3N_{(x)}$	Work	c Area
1.	1.51x10 ²¹ atoms							
2.			1.25mol					
3.						25 grams		

Honors Chemistry Stoichiometry Test Answers

Robert R. Redfield

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Tests Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services The ETS Test Collection Catalog Educational Testing Service. Test Collection, 1993 The major source of information on the availability of standardized tests Wilson Library BulletinCovers commercially available standardized tests and hard to locate research instruments The Science Teacher ,1982 Some issues are accompanied by a CD ROM on a selected topic The Pharmaceutical Era ,1905 A Whole Year of Chemistry Quizzes Siegfried Howell, 2019-12-12 This book A Whole Year of Chemistry Quizzes was written to provide easy to use and grade quizzes to assess the comprehension of honors students Advance Placement students AP and International Baccalaureate IB students This book of guizzes guides the teacher and the student through what is required in a non watered down chemistry course that leads students towards test and college readiness The outline of this book has a minimum of 4 guizzes per chapter that prepares students for the formative assessment associated at the end of all chapters The 25 chapters include topics that are covered in the honors chemistry setting as well as specialty topics like thermodynamics kinetics rates of reactions that are seen in the Advance Placement classes Included within this book are guizzes for the International Baccalaureate teacher that wishes to test students on environmental chemistry as well as biological and food chemistry. This is a book that was written to fill the void of valuable resources needed for novice and experienced teachers in institutions that continually push for more summative assessments higher DOKs and rapid feedback while limiting preparation time As a teacher for over 25 years I know that any well outlined structured and comprehensive resource saves time in additional planning searching and preparing Use this book to help you identify and test students on topics that are important to their comprehension and success with their final test Chapter 1 Matter and changeChapter 2 measurement and calculations Chapter 3 Atoms The building blocks of matterChapter 4 Arrangement of electrons in atomsChapter 5 The periodic lawChapter 6 Chemical bondingChapter 7 Chemical formulas and chemical compoundsChapter 8 Chemical equations and reactionsChapter 9 StoichiometryChapter 10 Physical characteristics of gasesChapter 11 Molecular composition of gasesChapter 12 Liquids and solidsChapter 13 SolutionsChapter 14 Ions in aqueous solution and colligative properties Chapter 15 Acids and bases Chapter 16 Acid base titrations Chapter 17 Reaction energy and reaction kineticsChapter 18 Chemical equilibriumChapter 19 Oxidation reduction reactionsChapter 20 Chemical thermodynamicsChapter 21 Carbon and hydrocarbonsChapter 22 Other organic compoundsChapter 23 Nuclear chemistryChapter 24 Biological and Food chemistryChapter 25 Environmental chemistry A Concrete Stoichiometry

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