

Problem 8: Modeling Hydrocarbon Equilibria

Problem Statement

Consider the problem of the following hydrocarbon system (see below). Using the Peng-Robinson and Soave-Redlich-Kwong equations of state, calculate the phase behavior of the system. Assume the hydrocarbon is a pure component. The system is at a temperature of 300 K and a pressure of 10 bar. The critical properties of the hydrocarbon are: $T_c = 300$ K, $P_c = 10$ bar, $Z_c = 0.27$.

- (a) Calculate the critical properties of the hydrocarbon using the Peng-Robinson equation of state.

- (b) Calculate the critical properties of the hydrocarbon using the Soave-Redlich-Kwong equation of state.

- (c) Calculate the critical properties of the hydrocarbon using the Peng-Robinson equation of state.

- (d) Calculate the critical properties of the hydrocarbon using the Soave-Redlich-Kwong equation of state.

- (e) Calculate the critical properties of the hydrocarbon using the Peng-Robinson equation of state.

Honors Chemistry Predicting Double Replacement Reactions

Tobias Bleicker

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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