

**Chapter Test B, continued**

- \_\_\_\_\_ 8. In a reaction, the ions of two compounds exchange places to produce another set of two ions and compounds. This reaction is called a \_\_\_\_\_.
- synthesis reaction
  - decomposition reaction
  - single-displacement reaction
  - double-displacement reaction
- \_\_\_\_\_ 9. The use of a double arrow in a chemical equation indicates that the reaction \_\_\_\_\_.
- is reversible
  - requires heat
  - is written backward
  - has not been confirmed in the laboratory

**PART II** Write the correct term (or terms) in the space provided.

10. A chemical equation is solved that begins with heat of a substance in a chemical equation is called  $\Delta H$ . \_\_\_\_\_
11. In a chemical equation, the symbol  $(g)$  indicates that the substance is \_\_\_\_\_.
12. The product is written for a single reactant as \_\_\_\_\_.
13. The general equation for a synthesis reaction is \_\_\_\_\_.
14. The general equation for a decomposition reaction is \_\_\_\_\_.
15. The general equation for a single-displacement reaction is \_\_\_\_\_.
16. The general equation for a double-displacement reaction is \_\_\_\_\_.

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