

Thermodynamics

MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question.

1. _____ If an ideal gas does positive work on its surroundings,
 - a. the temperature of the gas increases.
 - b. the volume of the gas increases.
 - c. the pressure of the gas increases.
 - d. the internal energy of the gas decreases.

2. _____ Sunlight strikes an ice cube at its melting point and causes it to melt. What is this process?
 - a. adiabatic process
 - b. isothermal process
 - c. isobaric process
 - d. isovolumetric process

3. _____ Which of the following processes for an ideal gas system has an unchanging internal energy and a heat intake that corresponds to the value of the work done by the system?
 - a. isothermal process
 - b. isobaric process
 - c. adiabatic process
 - d. isovolumetric process

4. _____ How is conservation of internal energy expressed for an isovolumetric system?
 - a. $Q = W = \Delta U = 0$; $U_i = U_f$
 - b. $Q = 0$; so $\Delta U = -W$
 - c. $\Delta T = 0$, so $\Delta U = Q - W = 0$, so $Q = W$
 - d. $\Delta V = 0$, so $P\Delta V = 0$ and $W = 0$, so $\Delta U = Q$

5. _____ A 4 mol ideal gas system undergoes an adiabatic process in which it expands and does 20 J of work on its environment. What is its change in internal energy?
 - a. -20 J
 - b. -5 J
 - c. 0 J
 - d. 20 J

6. _____ According to the second law of thermodynamics, the heat received by a heat engine operating in a complete cycle from a high-temperature reservoir
 - a. must be completely converted to work.
 - b. equals the entropy increase.
 - c. can be completely converted to internal energy.
 - d. cannot be completely converted to work.

Holt Physics Test Thermodynamics

JR Anderson



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Holt Physics Test Thermodynamics Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Holt Physics Test Thermodynamics**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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