

QUESTION 2

	Test	Observation	Inference
(a)(i)	C + Conc. H_2SO_4 + strong heat	A copious brown irritating fume which turned moist blue litmus paper to red	Gas is NO_2 NO_3^- present
(b)(i)	C + water + NaOH in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
(ii)	C + water + NH_3 in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
Trickle slowly down	(iii) C + water + freshly prepared $FeSO_4$ + Conc. H_2SO_4	A brown ring is formed at the junction of the acid and aqueous solution.	NO_3^-
(c)(i)	D + water	D dissolved	soluble
(ii)	Solution D + dil HCl	Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black	H_2S present
(iii)	Solution D + NaOH in drops, in excess	white precipitate precipitate insoluble	Ca^{2+} or Pb^{2+}
(iv)	solution D + $(NH_4)_2CO_3$	white precipitate	Ca^{2+}

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J. D. Stove, K. A. Phillips



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1. Understanding the eBook Gce 2013 Practical Chemistry Answers
 - The Rise of Digital Reading Gce 2013 Practical Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gce 2013 Practical Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gce 2013 Practical Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gce 2013 Practical Chemistry Answers
 - Personalized Recommendations
 - Gce 2013 Practical Chemistry Answers User Reviews and Ratings
 - Gce 2013 Practical Chemistry Answers and Bestseller Lists
5. Accessing Gce 2013 Practical Chemistry Answers Free and Paid eBooks
 - Gce 2013 Practical Chemistry Answers Public Domain eBooks
 - Gce 2013 Practical Chemistry Answers eBook Subscription Services
 - Gce 2013 Practical Chemistry Answers Budget-Friendly Options
6. Navigating Gce 2013 Practical Chemistry Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gce 2013 Practical Chemistry Answers Compatibility with Devices
 - Gce 2013 Practical Chemistry Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gce 2013 Practical Chemistry Answers
 - Highlighting and Note-Taking Gce 2013 Practical Chemistry Answers
 - Interactive Elements Gce 2013 Practical Chemistry Answers
8. Staying Engaged with Gce 2013 Practical Chemistry Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Gce 2013 Practical Chemistry Answers
- 9. Balancing eBooks and Physical Books Gce 2013 Practical Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gce 2013 Practical Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gce 2013 Practical Chemistry Answers
 - Setting Reading Goals Gce 2013 Practical Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gce 2013 Practical Chemistry Answers
 - Fact-Checking eBook Content of Gce 2013 Practical Chemistry Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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