Example: Determine the general rule  $(T_n)$  for this number pattern.

Now let's determine the general term of the linear b+c (5) pottern 5;7;9;11;..... using these filmdings.

$$B \approx 2$$

$$b+c=5$$
 (first term)

$$\therefore 2+c=5$$
 (substitute  $b=2$  into the equation  $b+c=5$ )

$$T_{\alpha} = \delta m + c$$

$$z_n T_n = 2m + 3$$

#### Answer the following questions:

List the following three numbers in each pattern

1. 10 m 3.

- 1.1 -4 -6 -8 -10 ....
- 1.2 7, 14, 21, 28.....
- 1.3 5.0, 11, 14.....
- 1.4 8, 5, 2, -1......



- How many small triangles will there be in pattern 4? 22.3
- 2.2 Determine the general rule  $(T_n)$  for this number pattern.
- How many small triangles will there be in pattern 13? 2.3

Consider the following number pattern:

- Write down the following two terms of this pattern. 236.75
- Determine the n th term for the terms in the uneven positions. 20.20
- 39.38 Which term will be the first negative term of this pattern?

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related to  $\Delta g$  for a process through eqn 1 and as long as  $\Delta h$  and  $\Delta s$  are independent of temperature eqn 1 can be combined with eqn 2 to determine  $\Delta h$  and  $\Delta s$  from the temperature dependence of k

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18 1 solubility product constant ksp chemistry libretexts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as ca 3 po 4 2 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore k ca 2 3 po 3 4 2 ca 3 po 4 2 ca 3 po 4 2 k ksp ca 2 3 po 3 4 2

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value tothe theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the ca 2 and oh concentrations in solution

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