

Рядный топливный насос высокого давления

Общие сведения

На двигателях QST30, независимо от их практического предназначения, используются топливные насосы высокого давления Bosch. На промышленных моделях двигателей используются модели TNEB RP39, показанные здесь.

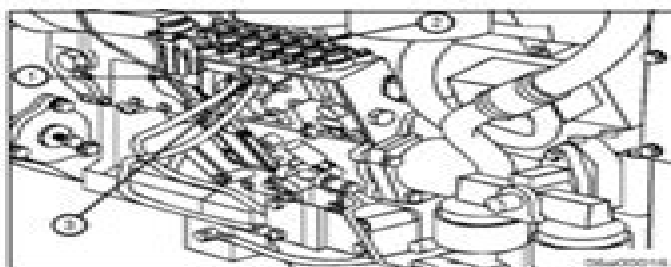
На двигателях модели G-Drive используются TNEB модели RP500 с фланцевым или открытым приводом.

Рядный топливный насос высокого давления выполняет три основных функции:

- Создает высокое давление топлива, необходимое для его впрыска.
- Дозирует точное количество топлива, необходимое для впрыска.
- Распределяет высокое давление и дозированное количество топлива к каждому цилиндру в определенный момент времени.

В рядном топливном насосе высокого давления используются отдельные плунжеры для создания и распределения высокого давления топлива, необходимого для впрыска.

Изношенный или поврежденный плунжер в рядном TNEB оказывает влияние на работу только одного цилиндра.



Снятие

Топливный насос высокого давления с фланцевым приводом

ПРИМЕЧАНИЕ: Для снятия правостороннего TNEB проверните коленчатый вал двигателя до метки синхронизации момента впрыска для данного двигателя, указанной в технической характеристике.

Установите защитную крышку, чтобы предотвратить попадание грязи и пыли в магнетальное окно TNEB или впускное окно соединителя форсунок. Очистите от загрязнения зону вокруг рядного TNEB.

Снимите электрические разъемы (1) и топливопроводы (2). Снимите все маслопроводы (3).

Hornet 698 Manual Rus

Sebastian Brünink

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Hornet 698 Manual Rus:

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mole moles of H_2O 40 18 2 2 moles therefore according to the equation

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time the total pressure $p_a \times \text{mole fraction}_a = \text{mole}_a / \text{moles total}$ right sum $n = \sum x_i$ mole fraction of substance
a is often denoted by x_a and

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web 1 in the given mixture the sum of all the mole fractions is equal to one $x_a + x_b + \dots = 1$ the multiplication of the mole fraction
by 100 gives the mole percentage 3 mole fraction is a unitless and dimensionless expression advantages of molar fraction
mole fraction is not dependent on the temperature

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web nov 6 2023 $x_a \text{ mol a} / \text{mol a} + x_b \text{ mol b} / \text{mol a} + \dots = 1$ note in the given mixture the sum of all the mole fractions
present is equal to one $x_a + x_b + \dots = 1$ when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is
a unitless and dimensionless expression image will be uploaded soon

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mass C_6H_6 6 12 6 1 g mol 1 78 g mol 1 hence C_6H_6 s number of moles 30 78 mol 0 3846 mol molar mass of carbon tet CCl_4 1
12 4 355 154 g mol 1

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g of water calculate the molarity and molality of the solution given $n = 14$ $h = 1$ $c = 12$ $o = 16$ given mass of solute urea 11 11 g mass
of solvent water 100 g 0 1 kg to find molarity of solution solution

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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