

Name: _____

Honors Chemistry Unit 1 Study Guide: Chapters 4, 11, & 19

1. What particles are found in the nucleus of an atom? What is the charge of the nucleus? What is the charge of the atom? It would be helpful to draw and label this.

In the nucleus, there are protons and neutrons, the charge is positive, a charge of an atom is neutral

2. Draw and label the parts found within a box on the periodic table.

3. How is an atom's atomic number related to its number of protons? Number of electrons?

The atom's number is the same as it's number of protons and electrons

4. What is the charge of the nucleus for element atomic number 56?

It is positive charge

5. Which particles account for most of an atom's mass?

The protons and neutrons

6. How is mass number related to the number of protons and neutrons an atom has?

The mass number is the sum of protons and neutrons

7. When you do not know the units of mass, what can you use? Write the full name and abbreviation.

You can use amu or atomic mass units

8. Distinguish between Democritus' idea and Dalton's atomic theory.

9. What parts of Dalton's atomic theory are now known to be false?

10. Draw and label the five atomic models discussed in class.

Honors Chemistry Study Guide

RJ Shavelson



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Table of Contents Honors Chemistry Study Guide

1. Understanding the eBook Honors Chemistry Study Guide
 - The Rise of Digital Reading Honors Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Honors Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Honors Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Honors Chemistry Study Guide
 - Personalized Recommendations
 - Honors Chemistry Study Guide User Reviews and Ratings
 - Honors Chemistry Study Guide and Bestseller Lists
5. Accessing Honors Chemistry Study Guide Free and Paid eBooks
 - Honors Chemistry Study Guide Public Domain eBooks
 - Honors Chemistry Study Guide eBook Subscription Services
 - Honors Chemistry Study Guide Budget-Friendly Options
6. Navigating Honors Chemistry Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Honors Chemistry Study Guide Compatibility with Devices
- Honors Chemistry Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Honors Chemistry Study Guide
 - Highlighting and Note-Taking Honors Chemistry Study Guide
 - Interactive Elements Honors Chemistry Study Guide
- 8. Staying Engaged with Honors Chemistry Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Honors Chemistry Study Guide
- 9. Balancing eBooks and Physical Books Honors Chemistry Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Honors Chemistry Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Honors Chemistry Study Guide
 - Setting Reading Goals Honors Chemistry Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Honors Chemistry Study Guide
 - Fact-Checking eBook Content of Honors Chemistry Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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