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1. Which of the following processes provide evidence of the particulate nature of matter?

I. Capillarity
II. Diffusion
III. Filtration
IV. Osmosis

(A) I and III only
(B) I and IV only
(C) II and III only
(D) II and IV only

2. 'Mass number' is the number of

(A) neutrons plus protons
(B) neutrons minus protons
(C) electrons plus neutrons
(D) electrons plus protons

3. Which of the following elements has 7 electrons in its outer shell?

(A) Hydrogen
(B) Oxygen
(C) Nitrogen
(D) Chlorine

4. An atom that has 20 electrons when it ionizes has the same electronic configuration as

(A) argon
(B) calcium
(C) neon
(D) sodium

5. $^{23}_{11}\text{Na}$ and $^{24}_{11}\text{Na}$ are isotopes because they

(A) conduct electricity
(B) react vigorously with water
(C) have the same number of protons
(D) have the same relative atomic mass

6. Sulfur and oxygen are in the same group of the periodic table because

(A) they can react with each other
(B) they can form covalent compounds
(C) they have the same number of electrons in their outer shell
(D) the atomic number of sulfur is 16 and the relative atomic mass of oxygen is 16

Item 7 refers to the following quantities of atoms.

I. 12 g of carbon
II. 16 g of sulfur
III. 23 g of sodium
IV. 12 g of magnesium

(Relative mass number: C = 12; S = 32; Mg = 24; Na = 23)

7. Which of the following pairs represents equal quantities of atoms?

(A) I and II only
(B) I and IV only
(C) II and III only
(D) II and IV only

8. An ion with a single negative charge may be converted into a neutral atom by

(A) losing a neutron
(B) gaining a proton
(C) losing an electron
(D) gaining an electron

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Exam Practice Workbook CGP Books, 2016-05-09 **GCSE Physics Edexcel 10 Minute Tests**, 2018-08-28 Edexcel GCSE Physics Ron Holt, 2008 *Model Answers* David F. Nuttall, 1994-01-01 **New Grade 9-1 Edexcel International**

GCSE Physics: 10-Minute Tests (with Answers) CGP Books, 2018-12-03 *GCSE Physics* CGP Books, Sarah Hilton, 2007-01-01 *GCSE Physics Workbook Including Answers* *Essential GCSE Physics* Lister Appleton Redman, 1988

Revision for Physics GCSE Ken Foulds, John Watts, 1999 This work is intended for students preparing for GCSE Science Physics or the physics component of GCSE Science Double Award and covers all GCSE Physics syllabuses including IGCSE Information is presented in manageable units and quick questions on each double page spread can be used as a test **Gcse Physics** Richard Parsons, 2004-02-18 These GCSE Physics Practice Papers for Higher level are specially designed for retail and offer fantastic value for parents and children everywhere They ll help pupils find out what sort of questions come up and how they should answer them The answer book s also got a proper mark scheme just like the real thing This book provides ideal last minute exam preparation **GCSE Edexcel Physics**, *GCSE Physics Higher* Brian Arnold, 2003 **Physics Higher** Letts GCSE, 2018-01-25 Exam Board AQA Level GCSE Grade 9 1 Subject Physics First Teaching September 2015 First Exams June 2017 Need more exam practice Letts will get you through your GCSE 9 1 exam Have a go at 2 complete tests Questions just like the real thing All the answers at the back **New Grade 9-1 GCSE Physics Exam Practice Workbook (with Answers)** CGP Books, 2016-05-03 *Questions for GCSE Physics* Nick England, David Harrison, 1997 This volume provides exam questions for GCSE physics students to use either as homework or for practice Each chapter deals with a particular topic and starts with a summary of important information and equations followed by worked examples Answers to selected questions are provided Exam Insights for GCSE Physics Ayd Instone, 2020 Improve your grade by focusing on the lessons learnt from previous exams This write in workbook targets the tricky topics in GCSE Physics with insights from the examiners reports model answers and practice questions to secure understanding and improve exam performance Target revision focusing on the topics and skills students have struggled with in previous exams Improve awareness of what the examiners are looking for and what s required to achieve each grade with model answers commentary and full mark schemes

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