

Key

Molecular Geometry

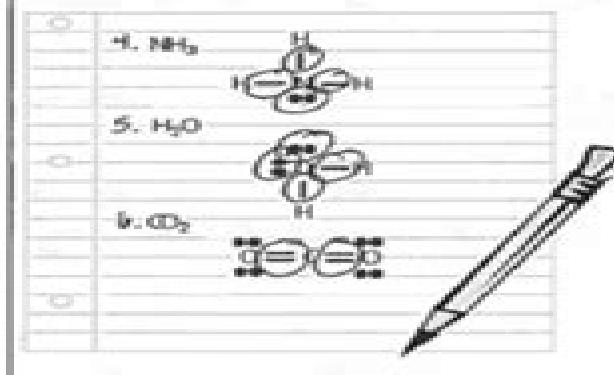
How can molecular shapes be predicted using the VSEPR theory?

Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are three-dimensional. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

Model 1 – Lewis Structures

Lewis Structures	
1. H ₂ CO	
2. PF ₃	
3. O ₃	



Lone pair = ::

H_2CO 3 electron domains (3 bonding, 0 nonbonding)	3-D Molecular Shape trigonal planar
BeF_3 2 electron domains (2 bonding, 0 nonbonding)	 linear
CH_4 4 electron domains (4 bonding, 0 nonbonding)	 tetrahedral
NH_3 4 electron domains (3 bonding, 1 nonbonding)	 pyramidal
H_2O 4 electron domains (2 bonding, 2 nonbonding)	 bent
CO_2 2 electron domains (2 bonding, 0 nonbonding)	 linear

Molecular Geometry Pogil Activities For High School

Open University Bletchley

Molecular Geometry Pogil Activities For High School:

Molecular Geometry Alison Rodger,Mark Rodger,2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry Chapter 1 reviews the definition and determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical changes Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements beyond the second row The book also talks about the complexes of transition metals and f block elements and then covers the organometallic compounds and transition metal clusters The last chapter tackles the consequences of small local variations in geometry The text will be of great use to chemists who primarily deal with the properties of molecules and atoms

The VSEPR Model of Molecular Geometry Ronald J. Gillespie,Istvan Hargittai,2012-01-01 Valence Shell Electron Pair Repulsion VSEPR theory is a simple technique for predicting the geometry of atomic centers in small molecules and molecular ions This authoritative reference written by the developer of VSEPR theory features extensive coverage of structural information as well as theory and applications Helpful data on molecular geometries bond lengths and band angles appear in tables and other graphics 1991 edition

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Shape in Chemistry Paul G. Mezey,1993 **Misconceptions in the Area of Molecular Geometry and Molecular Polarity Among College Introductory and General Chemistry Students** Patricia L. Rambo,2004 Determination of Molecular Geometry by Spectroscopic Methods ,1972 Methods for Improved

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