

Name _____

Molecular Geometries Lab Report Sheet

Formula	Lewis Structure	No of electron domains/lig	Electron domain geometry	VSEPR Molecular geometry	Hybridization of central atom	Polar ?
CO_2	$\text{O}=\text{C}=\text{O}$ 24 valence 24 2 charge clouds	0	Linear 180°	Linear 	sp	NP
BF_3	$\text{F}-\text{B}-\text{F}$ 24 32 3 charge clouds	0	Trigonal planar 120°		sp^2	NP
SnCl_2	$\text{Cl}-\text{Sn}-\text{Cl}$ 16 24	3	Bent <120°		sp^2	P
BF_4^-	$\left[\text{F}-\text{B}-\text{F} \right]^-$ 24 40	0	Tetrahedral 109.5°		sp^3	P
AsH_3	$\text{H}-\text{As}-\text{H}$ 8	4	Trigonal pyramidal 109.5°		sp^3	P
H_2S	$\text{H}-\text{S}-\text{H}$ 8	2	Bent 109.5°		sp^3	NP
NH_2^-	$\left[\text{H}-\text{N}-\text{H} \right]^-$ 8	2	Bent 109.5°		sp^3	P
POF_3	$\text{F}-\text{P}-\text{Cl}$ 10 24	0	Trigonal bipyre 120° + 90°		sp^3d	P

Molecular Geometry Lab Answers

Brendan G. Carr



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