# **Mole Calculations**

# Multiple Choice Questions

1

How many moles are there in 6 dm3 of oxygen at r.t.p.?

A 0.25 moles

B 0.50 moles

C 0.60 moles

D 3.00 moles

2

What is the total volume of gas remaining after 20 cm<sup>3</sup> ethane are burned completely in 100 cm3 oxygen? All volumes are measured at the same pressure and the same temperature, which is above 100 °C.

C<sub>2</sub>H<sub>6</sub> + 3 ½ O<sub>2</sub> → 2CO<sub>2</sub> + 3H<sub>2</sub>O

A 40 cm<sup>3</sup>

B 100 cm<sup>3</sup>

C 120 cm<sup>3</sup>

D 130 cm<sup>3</sup>

3

What is the mass of 0.36 moles of ethanoic acid, CH<sub>3</sub>COOH?

A 0.06g 🔘

B 21.60g (

C 6.00g 🔘

D 36.00g (

# **Mole Practice Problems**

Barron's Educational Series, Mark Kernion, Joseph A. Mascetta

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by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields Comprehensive Chemistry XI Dr. B. Kapila, S. K. Khanna, 2010-11 Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education CBSE ICSE Chemistry for Class X (A.Y. 2023-24)Onward V. K. Sally, 2023-05-20 A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination New Delhi for Class X The main strength of the book lies in the presentation of scientific content which has been so arranged that the topics are linked with each other and do not cause any unnecessary burden on the mind of the student Salient features of this book are as follows Clear simple and easy language A large number of chemical reactions along with experimental observations Full sized diagrams as expected from a student in the examination Topic wise Video Lectures as a support for effective learning Concepts supplemented by suitable day to day examples Periodic Table showing mass number atomic number of various elements along with the list of actual names of the elements Highlighted important terms and definitions Well designed exercises to assess conceptual reasoning skill and application based learning as per the latest recommendations of ICSE board Solved numerical problems at the end of each chapter to help the students solve numericals on their own A chapter on Practical Chemistry to help students in their laboratory work The latest Solved ICSE Specimen Paper has been given Scan QR codes given at the end of each chapter to get the solution of chapter wise ICSE Board Examination Questions We hope this book will prove useful to fellow teachers and students Suggestions for further improvement of this book shall be gratefully acknowledged Author Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

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