

CHEMISTRY MOLE COOKIES

Practice
mole
calculations
with a
cookie
recipe!



Chemistry Mole Cookies Worksheet

Instructions to complete the recipe for chocolate chip cookies:

1. Fill a 100 mL beaker halfway with water (ingredient's molecular mass).
2. Convert the ingredient amounts into mole units. Express water in grams, then grams to the proper decimal units. (Use the conversions below.)
3. Complete the recipe card on this sheet.

Note: Many of the ingredients are molecules, so have complete structural formulas. They have been supplied here.

INGREDIENT	MOLECULAR MASS <small>molecular mass</small>	MOLES <small>mole</small>	GRAMS <small>g</small>	COOKING UNIT <small>amount</small>
flour <chem>C6H12O6</chem>		1.00 moles		
baking soda <chem>NaHCO3</chem>		0.0001 moles		
salt <chem>NaCl</chem>		0.001 moles		
butter <chem>C4H8O2</chem>		1.000 moles		
white sugar <chem>C12H22O11</chem>		0.001 moles		
brown sugar <chem>C12H22O11</chem>		0.001 moles		
egg <chem>C6H12O6</chem>		0.001 moles		
vanilla <chem>C10H12O2</chem>		0.001 moles		
chocolate chips <chem>C12H18O6</chem>		1.00 moles		

Use the following conversions to get from grams to standard cooking units of amount.

- 1 cup flour = 125.00 grams
- 1 cup brown sugar = 225.00 grams
- 1 large egg = 50 grams
- 3 teaspoons vanilla = 4.75 grams
- 1 cup chocolate chips = 175.00 grams

- 1 cup white sugar = 200.00 grams
- 1 teaspoon baking soda = 0.001 grams
- 1 teaspoon salt = 0.001 grams
- 1 cup butter = 225.00 grams

continued on next

Mole Cookies Answers

RJ Alexander



Mole Cookies Answers:

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