

Molarity Calculations

Calculate the molarities of the following solutions:



- 1) 2.3 moles of sodium chloride in 0.45 liters of solution.

$$M = \frac{2.3 \text{ moles}}{0.45 \text{ L}} = 5.111 \frac{\text{mol}}{\text{L}} = \boxed{5.11 \text{ M}}$$



- 2) 1.2 moles of calcium carbonate in 1.22 liters of solution.

$$M = \frac{1.2 \text{ mol}}{1.22 \text{ L}} = 0.983606557 \frac{\text{mol}}{\text{L}} = \boxed{0.98 \text{ M}}$$



- 3) 0.09 moles of sodium sulfate in 12 mL of solution.

12 mL = 0.012 L

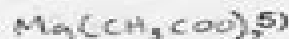
$$M = \frac{0.09 \text{ mol}}{0.012 \text{ L}} = \boxed{7.5 \text{ M}}$$



- 4) 0.75 moles of lithium fluoride in 65 mL of solution.

65 mL = 0.065 L

$$M = \frac{0.75 \text{ mol}}{0.065 \text{ L}} = 11.53846154 \text{ M} = \boxed{11.54 \text{ M}}$$



- 5) 0.8 moles of magnesium acetate in 5 liters of solution.

$$M = \frac{0.8 \text{ mol}}{5 \text{ L}} = \boxed{0.16 \text{ M}}$$



- 6) 120 grams of calcium nitrite in 240 mL of solution.

240 mL = 0.240 L

$$n = 120 \text{ g} \left(\frac{1 \text{ mol}}{132.09 \text{ g}} \right)$$

MW_{Ca(NO₂)₂} = 132.09 g/mol

n = 0.91 mol

$$M = \frac{0.91 \text{ mol}}{0.240 \text{ L}} = \boxed{3.79 \text{ M}}$$



- 7) 98 grams of sodium hydroxide in 2.2 liters of solution.

MW_{NaOH} = 39.997 g/mol

$$n = 98 \text{ g} \left(\frac{1 \text{ mol}}{39.997 \text{ g}} \right) = 2.45 \text{ mol}$$

$$M = \frac{2.45 \text{ mol}}{2.2 \text{ L}} = \boxed{1.11 \text{ M}}$$



- 8) 1.2 grams of hydrochloric acid in 25 mL of solution.

MW_{HCl} = 36.461 g/mol

$$n = (1.2 \text{ g}) \left(\frac{1 \text{ mol}}{36.461 \text{ g}} \right) = 0.033 \text{ mol}$$

$$M = \frac{0.033 \text{ mol}}{0.025 \text{ L}} = \boxed{1.32 \text{ M}}$$



- 9) 45 grams of ammonia in 0.75 L of solution.

MW_{NH₃} = 17.031 g/mol

$$n = 45 \text{ g} \left(\frac{1 \text{ mol}}{17.031 \text{ g}} \right) = 2.64 \text{ mol}$$

$$M = \frac{2.64 \text{ mol}}{0.75 \text{ L}} = \boxed{3.52 \text{ M}}$$

Molarity Calculations Chemfiesta

Marco Cascella

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Molarity Calculations Chemfiesta:

This book delves into Molarity Calculations Chemfiesta. Molarity Calculations Chemfiesta is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Molarity Calculations Chemfiesta, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Molarity Calculations Chemfiesta
- Chapter 2: Essential Elements of Molarity Calculations Chemfiesta
- Chapter 3: Molarity Calculations Chemfiesta in Everyday Life
- Chapter 4: Molarity Calculations Chemfiesta in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Molarity Calculations Chemfiesta. This chapter will explore what Molarity Calculations Chemfiesta is, why Molarity Calculations Chemfiesta is vital, and how to effectively learn about Molarity Calculations Chemfiesta.
3. In chapter 2, this book will delve into the foundational concepts of Molarity Calculations Chemfiesta. The second chapter will elucidate the essential principles that must be understood to grasp Molarity Calculations Chemfiesta in its entirety.
4. In chapter 3, this book will examine the practical applications of Molarity Calculations Chemfiesta in daily life. The third chapter will showcase real-world examples of how Molarity Calculations Chemfiesta can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Molarity Calculations Chemfiesta in specific contexts. The fourth chapter will explore how Molarity Calculations Chemfiesta is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Molarity Calculations Chemfiesta. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Molarity Calculations Chemfiesta.

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