

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write the full name (including the name of the unit) of the person that first gave the names to the following chemical elements and compounds.

1. The compound of a diatomic molecule with a molecular weight of 170.
  - a. gas
  - b. gas liquid
  - c. crystalline substance
  - d. liquid
2. The element of a diatomic molecule is a gas at room temperature.
  - a. nitrogen
  - b. hydrogen
  - c. chlorine-like particles
  - d. mixture of particles
3. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
4. The compound of a diatomic molecule is a gas at room temperature.
  - a. crystalline substance
  - b. crystalline substance
  - c. crystalline substance
  - d. crystalline substance
5. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
6. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
7. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
8. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
9. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture
10. The compound of a diatomic molecule is a gas at room temperature.
  - a. liquid
  - b. dissolved in water
  - c. solid
  - d. mixture

# Modern Chemistry Chapter 10 Test B Answers

**Sebastian Brünink**



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