

Acids & Bases Review

1. Name the following compounds as acids and circle which are weak acids:



nitric acid



sulfuric acid



hydrofluoric acid



carbonic acid



acetic acid



phosphoric acid

2. Write the formulas for the following bases:

a) calcium hydroxide



b) aluminum hydroxide



c) ammonium hydroxide



d) lithium hydroxide



3. Complete the following table:

	Acids	Bases
Definition	A substance that increases the concentration of H^+ (or H_3O^+) when added to water. or A substance that acts as a proton (H^+) donor.	A substance that increases the concentration of OH^- when added to water. or A substance that acts as a proton (H^+) acceptor.
Electrolyte in aqueous solution?	H^+	OH^-
pH	less than 7	greater than 7
Taste	sour	bitter
Litmus turns	blue turns red	red turns blue
Which reacts with metals to form H_2 gas?	yes	no

Conjugate acid-base pairs:

4. Identify the acid, the base, the conjugate acid and the conjugate base in the reactions:



5. An acid will form the conjugate base and a base will form the conjugate acid.

Modern Chemistry Acids And Bases Test Answers

JE Gale



Modern Chemistry Acids And Bases Test Answers:

Decoding **Modern Chemistry Acids And Bases Test Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Modern Chemistry Acids And Bases Test Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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