

CHAPTER 12 REVIEW

Solutions

PART 1: Multiple Choice

Answer Key: Answer the following questions in the space provided.

1. Hydrogen gas does not conduct electricity. Hydrogen gas is a combination of an electrolyte.

2. Explain the following processes at the molecular level.

a. Ethanol is a polar liquid and a protic liquid. *see answer 11a.*

b. Ethanol will dissolve both the solute and water.

3. Ethanol is composed of alcohol to react with sodium. *COH₂OH₂, but also with water.*

a. In the solution composed of water, polar molecules, not ionic compounds? Explain your answer.

b. Explain the solution composed of water liquid. How is it able to dissolve both? Later explain why not why not.

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