

**Assessment****Chapter Test A****Chapter: Acids and Bases**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?  
a. a sour taste  
b. a slippery feel  
c. the ability to conduct an electric current  
d. reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?  
a. HCl  
b. H<sub>2</sub>SO<sub>4</sub>  
c. H<sub>3</sub>PO<sub>4</sub>  
d. HNO<sub>3</sub>
- \_\_\_\_\_ 3. Acetic acid is found in  
a. vinegar.  
b. the stomach.  
c. antacids.  
d. oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it  
a. is an electrolyte.  
b. tastes bitter.  
c. dissociates to form OH<sup>-</sup> ions.  
d. All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?  
a. HCl  
b. HCO<sub>3</sub><sup>-</sup>  
c. H<sub>3</sub>O<sup>+</sup>  
d. H<sub>3</sub>PO<sub>4</sub>
- \_\_\_\_\_ 6. In the reaction represented by the equation  
$$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
 which of these is a conjugate acid-base pair?  
a. H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sub>2</sub>(aq)  
b. H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)  
c. CH<sub>3</sub>NH<sub>2</sub>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)  
d. HC<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)

# Modern Chemistry Chapter 5 Practice Test

**Thorpe**

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