

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Chapter Test A

## Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
  - a slippery feel
  - the ability to conduct an electric current
  - reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?
- HCl
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_3\text{PO}_4$
  - $\text{HNO}_3$
- \_\_\_\_\_ 3. Acetic acid is found in
- vinegar.
  - the stomach.
  - antacids.
  - oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
  - tastes bitter.
  - dissociates to form  $\text{OH}^-$  ions.
  - All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?
- HCl
  - $\text{HCO}_3^-$
  - $\text{H}_3\text{O}^+$
  - $\text{H}_3\text{PO}_4$
- \_\_\_\_\_ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(aq) + \text{CH}_3\text{NH}_2(aq) \rightleftharpoons \text{HC}_2\text{O}_4^-(aq) + \text{CH}_3\text{NH}_3^+(aq),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_2(aq)$
  - $\text{H}_2\text{C}_2\text{O}_4(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{CH}_3\text{NH}_2(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$
  - $\text{HC}_2\text{O}_4^-(aq)$  and  $\text{CH}_3\text{NH}_3^+(aq)$

# Modern Chemistry Chapter 11 Test Answer Key

**R.A. Mackay, W. Henderson**



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**Chemistry** John A. Olmsted,Gregory M. Williams,Robert Charles Burk,2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI

units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant **Elementary Modern**

**Chemistry** Wilhelm Ostwald, Harry Wheeler Morse, 1909 **Essentials of Modern Chemistry** Charles Elwood Dull, 1918

*Introduction to Modern Inorganic Chemistry, 6th edition* R.A. Mackay, W. Henderson, 2017-12-21 This popular and comprehensive textbook provides all the basic information on inorganic chemistry that undergraduates need to know For this sixth edition the contents have undergone a complete revision to reflect progress in areas of research new and modified techniques and their applications and use of software packages *Introduction to Modern Inorganic Chemistry* begins by explaining the electronic structure and properties of atoms then describes the principles of bonding in diatomic and polyatomic covalent molecules the solid state and solution chemistry Further on in the book the general properties of the periodic table are studied along with specific elements and groups such as hydrogen the s elements the lanthanides the actinides the transition metals and the p block Simple and advanced examples are mixed throughout to increase the depth of students understanding This edition has a completely new layout including revised artwork case study boxes technical notes and examples All of the problems have been revised and extended and include notes to assist with approaches and solutions It is an excellent tool to help students see how inorganic chemistry applies to medicine the environment and biological topics

*General Chemistry* Ralph H. Petrucci, 1989 Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters, 1902 Comprehensive Analytical Chemistry Cecil Leeburn Wilson, David Woodburn Wilson, 1959 Books in Print Supplement , 2002 Technical Abstract Bulletin , 1978 Tanning Chemistry Anthony D Covington, 2015-11-09

Even in the 21st Century the manufacture of leather retains an air of the dark arts still somewhat shrouded in the mysteries of a millennia old craft based industry Despite the best efforts of a few scientists over the last century or so much of the understanding of the principles of tanning is still based on received wisdom and experience Leather is made from usually the hides and skins of animals large animals such as cattle have hides small animals such as sheep have skins The skin of any animal is largely composed of the protein collagen so it is the chemistry of this fibrous protein and the properties it confers to the skin with which the tanner is most concerned In addition other components of the skin impact on processing impact on the chemistry of the material and impact on the properties of the product leather Therefore it is useful to understand the relationships between skin structure at the molecular and macro levels the changes imposed by modifying the chemistry of the material and the eventual properties of the leather This book aims to contribute to changing the thinking in the industry to continue building a body of scientific understanding aimed at enhancing the sustainability of an industry which produces a unique group of materials derived from a natural source The Science of Leather is the only current text on tanning science

and addresses the scientific principles which underpin the processes involved in making leather It is concerned with the chemical modification of collagen prior to tanning and the tanning reactions in particular The subject is covered in the following order collagen chemistry collagen structure skin structure processing to prepare for tanning the tanning processes and processing after tanning The aim of the book is to provide leather scientists and technologists with an understanding of how the reactions work the nature of their outcomes and how the processes can be controlled and changed The objective is to synthesise a scientific view of leather making and to arrive at an understanding of the nature of tanning how the wide range of chemistries employed in the art can change the properties of collagen making leather with different properties especially conferring different degrees of stabilisation as measured by the hydrothermal stability Environmental issues are not treated as a separate theme the impact of leather making on the environment is a thread running through the text with the assumption that better understanding of the science of leather making will lead to improved processing The book also reflects on the ways leather technology may develop in the future based on the foundation of understanding the scientific principles which can be exploited It also includes a subject index references and a glossary The book provides the reader with insights into the role science plays in leather technology and provides fundamental understanding which should be the basis for scientific and technological research and development for the benefit of the global leather industry The book is aimed at students leather scientists and technologists in both academia and industry in leather production and in chemical supply houses

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 Nicholas D. Tzimopoulos, 1990     **Modern Urinology** Clifford Mitchell, 1912     *Basic Concepts of Chemistry* Leo J. Malone, Theodore Dolter, 2008-12-03. Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

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