

(iii) Write the equation for the termination step which produces a hydrocarbon as a product in this reaction.

(1)

(b) Ethene also reacts with chlorine but by a different mechanism.

* (i) Describe how the π bond in ethene forms and explain why this bond causes ethene to take part in addition reactions with halogens.

(2)

* (ii) Write the mechanism for the reaction of ethene with chlorine.

Use curly arrows to show movements of electron pairs.

(3)



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Table of Contents June 2ial Edexcel C3question Papers

1. Understanding the eBook June 2ial Edexcel C3question Papers
 - The Rise of Digital Reading June 2ial Edexcel C3question Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying June 2ial Edexcel C3question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 2ial Edexcel C3question Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 2ial Edexcel C3question Papers

- Personalized Recommendations
- June 2ial Edexcel C3question Papers User Reviews and Ratings
- June 2ial Edexcel C3question Papers and Bestseller Lists
- 5. Accessing June 2ial Edexcel C3question Papers Free and Paid eBooks
 - June 2ial Edexcel C3question Papers Public Domain eBooks
 - June 2ial Edexcel C3question Papers eBook Subscription Services
 - June 2ial Edexcel C3question Papers Budget-Friendly Options
- 6. Navigating June 2ial Edexcel C3question Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - June 2ial Edexcel C3question Papers Compatibility with Devices
 - June 2ial Edexcel C3question Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 2ial Edexcel C3question Papers
 - Highlighting and Note-Taking June 2ial Edexcel C3question Papers
 - Interactive Elements June 2ial Edexcel C3question Papers
- 8. Staying Engaged with June 2ial Edexcel C3question Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers June 2ial Edexcel C3question Papers
- 9. Balancing eBooks and Physical Books June 2ial Edexcel C3question Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 2ial Edexcel C3question Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine June 2ial Edexcel C3question Papers
 - Setting Reading Goals June 2ial Edexcel C3question Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of June 2ial Edexcel C3question Papers

- Fact-Checking eBook Content of June 2ial Edexcel C3question Papers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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