

# Acids and Bases

How do acids and bases behave in water?

## Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

## Model 1 – Arrhenius Acids and Bases

| Common Name for Aqueous Solution | Chemical Formula                                      | Found in...       | Tastes... | Turns Litmus Paper... | Conducts Electricity? | Acid or Base |
|----------------------------------|-------------------------------------------------------|-------------------|-----------|-----------------------|-----------------------|--------------|
| Acetic acid                      | $\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$          | Vinegar           | Sour      | Red                   | Yes                   | Acid         |
| Benzoic acid                     | $\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$          | Food preservative | Sour      | Red                   | Yes                   | Acid         |
| Phosphoric acid                  | $\text{H}_3\text{PO}_4(\text{aq})$                    | Soda pop          | Sour      | Red                   | Yes                   | Acid         |
| Hydrochloric acid                | $\text{HCl}(\text{aq})$                               | Stomach acid      | Sour      | Red                   | Yes                   | Acid         |
| Citric acid                      | $\text{H}_2\text{C}_6\text{H}_7\text{O}_7(\text{aq})$ | Citrus fruits     | Sour      | Red                   | Yes                   | Acid         |
| Ascorbic acid                    | $\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$ | Vitamin C         | Sour      | Red                   | Yes                   | Acid         |
| Magnesium hydroxide              | $\text{Mg}(\text{OH})_2(\text{aq})$                   | Milk of magnesia  | Bitter    | Blue                  | Yes                   | Base         |
| Aluminum hydroxide               | $\text{Al}(\text{OH})_3(\text{aq})$                   | Antacids          | Bitter    | Blue                  | Yes                   | Base         |
| Barium hydroxide                 | $\text{Ba}(\text{OH})_2(\text{aq})$                   | Lubricants        | POISON    | Blue                  | Yes                   | Base         |
| Sodium hydroxide                 | $\text{NaOH}(\text{aq})$                              | Drain cleaner     | POISON    | Blue                  | Yes                   | Base         |

- Refer to Model 1.
  - What is the common chemical name for Vitamin C?  
**ascorbic acid**
  - Is Vitamin C classified as an acid or a base?  
**acid**
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.  
**sour, red, conduct electricity**
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

# Introduction To Acids Bases Pogil Answers

**Adams**



## Introduction To Acids Bases Pogil Answers:

*POGIL* Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

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## **Table of Contents Introduction To Acids Bases Pogil Answers**

1. Understanding the eBook Introduction To Acids Bases Pogil Answers
  - The Rise of Digital Reading Introduction To Acids Bases Pogil Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Introduction To Acids Bases Pogil Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Introduction To Acids Bases Pogil Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Introduction To Acids Bases Pogil Answers
  - Personalized Recommendations
  - Introduction To Acids Bases Pogil Answers User Reviews and Ratings
  - Introduction To Acids Bases Pogil Answers and Bestseller Lists
5. Accessing Introduction To Acids Bases Pogil Answers Free and Paid eBooks

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  - Introduction To Acids Bases Pogil Answers eBook Subscription Services
  - Introduction To Acids Bases Pogil Answers Budget-Friendly Options
6. Navigating Introduction To Acids Bases Pogil Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Introduction To Acids Bases Pogil Answers Compatibility with Devices
    - Introduction To Acids Bases Pogil Answers Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Introduction To Acids Bases Pogil Answers
    - Highlighting and Note-Taking Introduction To Acids Bases Pogil Answers
    - Interactive Elements Introduction To Acids Bases Pogil Answers
  8. Staying Engaged with Introduction To Acids Bases Pogil Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Introduction To Acids Bases Pogil Answers
  9. Balancing eBooks and Physical Books Introduction To Acids Bases Pogil Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Introduction To Acids Bases Pogil Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Introduction To Acids Bases Pogil Answers
    - Setting Reading Goals Introduction To Acids Bases Pogil Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Introduction To Acids Bases Pogil Answers
    - Fact-Checking eBook Content of Introduction To Acids Bases Pogil Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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