# Answer Key

#### Chapter 1

#### Lesson 1

#### Page I, Teaching

The complete subjects appear on the first answer lines; the complete predicates appear on the second answer lines.

- T. The sun
- is much bigger than all the other planets
- Planets in the solar system reflect light from the sun
- 3. Stars.
- twinkle in the sky at night.
- 4. Planets
  - shine with a steadler light
- Stars such as our sun make their own light and heat.
- 6. Some of the planets
- have satellites, or moons
- Our planet has only one satellite
- 8. The largest planet
- is Jupiter
- Asteroids, meteoroids, and comets belong to our solar system
- 10. Many of the asteroids
  - orbit in an area between Mars and Jupiter
- The surface of Mercury is covered with many craters
- Masses of clouds cover the surface of Venus
- 13. Large amounts of iron in its rocks give Mars a reddish color
- 14. One of the moons of Jupiter
  - is named Ganymede
- 15. Ganymede
  - is the biggest satellite in our solar system
- 16. The rings of Saturn
- can be seen through a telescope
- 17. The Voyager 2 mission
- discovered ten new moons of Uranus 18. Voyager
- sent scientists new information about Neptune
- Many astronomers use telescopes for studying the universe.
- 20. Some astronomers
- analyze light and radio waves 21. The Hubble Space Telescope
  - sends scientists new information daily
- 22. The planets closest to Earth are Mercury, Venus, and Mars

#### Page 2. More Practice

#### Exercise A.

The complete subjects appear on the first answer lines; the complete predicates appear on the second answer lines.

- 1. Meteors
  - sometimes enter Earth's atmosphere
- 2. Friction with the air
  - makes the meteors glow
- Glowing meteors are called falling stars
- 4. Some fast meteors
- travel about 26 miles per second
- 5. Large swarms of meteors enter
- Earth's atmosphere several times each year
- The shower of sparks from the meteors appears in the sky
- People throughout Earth's history have seen meteor showers
- 8. A meteorise
  - is a meteor that reaches the surface of the earth
- 9. Some meteorites
  - cause huge craters in Earth's surface
- The largest meteorite now on Earth weighs about 66 tons

#### Exercise I

Answers will vary. Possible answers are provided.

- 1. CP; My brother set the table for dinner.
- 2. CS: A newborn kitten is soft and cute.
- 3. CS; Ghost stories keep me awake at night.
- 4. CP: The swimmer broke the school record.
- 5. CP; Natalie gave her friend a gift.

#### Page 3, Application

#### Exercise A.

Answers will vary. Sample answers are provided.

- 1. Planets move around the sun.
- 2. Stars twinkle at night.
- 3. Spaceships travel to distant planets.
- 4. Astronauts in the space program train hard.
- 5. Satellites from Earth circle the planet.
- 6. The universe expands all the time.

#### Exercise B.

Answers will vary. A possible answer is provided. Asteroids are made of dark, rocky matter.

Asteroids have different sizes and shapes. They revolve around the sun, mainly between the orbits of Mars and Jupiter. Thousands of asteroids are known to astronomers. Ceres, the largest asteroid, was the first one to be discovered. It was identified on January 1, 1801. Other asteroids are named Pallas, Juno, and Vesta. Sometimes asteroids collide with each other. Pieces that fall to Earth are called meteorites. New asteroids are still being discovered.

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web respond to the questions and prompts in the orange boxes atomic radius electron affinity electron cloud energy level

group ion ionization energy metal nonmetal nucleus period periodic trends picometer valence electron do these before using the gizmo

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web jun 30 2023 major periodic trends include electronegativity ionization energy electron affinity atomic radius melting point and metallic character periodic trends arising from the arrangement of the periodic table provide chemists with an invaluable tool to quickly predict an element s properties

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web jan 28 2019 by mapping specific trends directly onto a blank periodic table the activity uses the periodic table as a foundational aspect of the major trends allowing for students to make connections between placement on the periodic table and a specific trend

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i 1 900 kj mol i 2 1757 kj mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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web the purpose of this lab is to introduce students to periodic trends it gives students an opportunity to explore the values of different properties for all elements in the periodic table to determine which properties of the elements follow a periodic trend

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web 1 name date per periodic trends lab periodic law states that when the elements are arranged in order of increasing atomic number there is a periodic repetition of their chemical and physical properties elements within the same group have the most similar chemical and physical properties although not identical

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web 1 rank the following elements by increasing atomic radius carbon aluminum oxygen potassium oxygen carbon aluminum

potassium 2 rank the following elements by increasing electronegativity sulfur oxygen neon aluminum neon aluminum sulfur oxygen 3 why does fluorine have a higher ionization energy than iodine

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web simulation periodic trends answer key background in this investigation you will examine several periodic trends including atomic radius ionization energy and ionic radius you will

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below dry lab periodic trends answer key name studocu - Aug 07 2023

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web caragabriel preview chemistry test oct 7 11 43 terms misterlominyo preview terms in this set 25 each element has 3 numbers listed under it which value represents the atomic radius the first number below the diagram what are the units for the atomic radius pm picometers 10 negative twelve meters

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web mar 13 2023 across the periodic table sizes of atoms show the following trends with many irregularities size increases down a group the outermost electrons are in successively more extensive orbitals as n increases

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web similarities and differences among the elements give rise to so called periodic trends both across rows and within columns of the periodic table recognizing periodic trends in the physical and chemical properties of the elements is key to understanding the full value of the periodic table concepts periodic table periodic trends

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web question periodic trends straw lab the purpose of this lab is to create a visual representation of trends on the periodic table you will need answer the following questions and write a formal lab report use the information below to belp guide you through writing your lab report 1