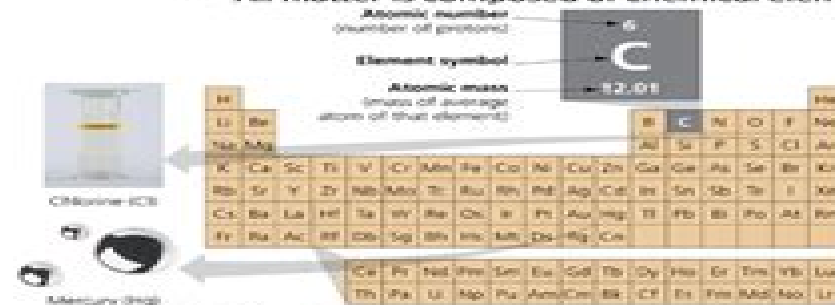
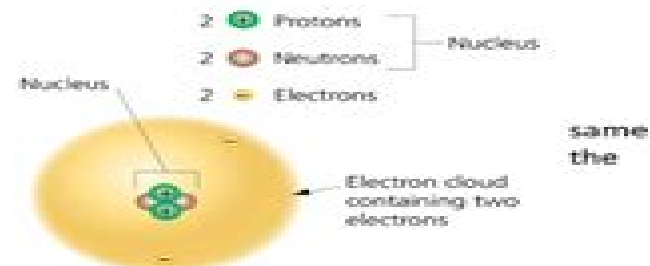


Chapter 2 Bio Essential Chemistry For Biology

- Take any biological system apart, and you eventually end up at the chemical level.
 - Matter** is anything that occupies space and has mass.
 - Mass** is a measure of the amount of material in an object.
 - An **element** is a substance that cannot be broken down into other substances by chemical reactions.
 - All matter is composed of chemical elements.



- Of the 92 naturally occurring elements, 25 are essential to people.
 - Four of these elements make up about 96% of the weight of the body.
 - Trace elements** are required in only very small amounts and are essential for life.
 - Elements can combine to form **compounds**, substances that contain two or more elements in a fixed ratio.
- Each element consists of one kind of atom.
- An **atom** is the smallest unit of matter that still retains the properties of an element.
- Atoms are composed of subatomic particles.
 - A **proton** is positively charged.
 - An **electron** is negatively charged.
 - A **neutron** is electrically neutral.
- All atoms of a particular element have the unique number of protons. This number is element's **atomic number**.
- An atom's **mass number** is the sum of the number of protons and neutrons in its nucleus.
 - Isotopes** have the same number of protons and behave identically in chemical reactions, but they have different numbers of neutrons.
 - A **radioactive isotope** is one in which the nucleus decays spontaneously.
- Only electrons are directly involved in chemical reactions.



Essential Experiments For Chemistry Answer Key

Lin, Cheng-Yao, Sun, Li



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Resources in Education ,1997

Zoology 1 Alfred Marlyn Elliott,1957

Insights from Research in Science

Teaching and Learning Nicos Papadouris,Angela Hadjigeorgiou,Constantinos P. Constantinou,2015-11-02 This book includes studies that represent the state of the art in science education research and convey a sense of the variation in educational traditions around the world The papers are organized into six main sections science teaching processes conceptual understanding reasoning strategies early years science education and affective and social aspects of science teaching and learning The volume features 18 papers selected from the most outstanding papers presented during the 10th European Science Education Research Association ESERA Conference held in Nicosia Cyprus in September 2013 The theme of the conference was Science Education Research for Evidence based Teaching and Coherence in Learning The studies presented underline aspects of great relevance in contemporary science education the need to reflect on different approaches to enhance our knowledge of learning processes and the role of context designed or circumstantial formal or non formal in learning and instruction These studies are innovative in the issues they explore the methods they use or the ways in

which emergent knowledge in the field is represented The book is of interest to science educators and science education researchers with a commitment to evidence informed teaching and learning

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