

45) How many moles of sodium carbonate contain  $1.773 \times 10^{17}$  carbon atoms?

- A)  $5.890 \times 10^{-7}$
- B)  $2.945 \times 10^{-7}$
- C)  $1.473 \times 10^{-7}$
- D)  $8.836 \times 10^{-7}$
- E)  $9.817 \times 10^{-8}$

46) How many grams of sodium carbonate contain  $1.773 \times 10^{17}$  carbon atoms?

- A)  $3.121 \times 10^{-5}$
- B)  $1.011 \times 10^{-5}$
- C)  $1.517 \times 10^{-5}$
- D)  $9.100 \times 10^{-5}$
- E)  $6.066 \times 10^{-5}$

47) The compound responsible for the characteristic smell of garlic is allicin,  $C_6H_{10}OS_2$ . The mass of 1.00 mol of allicin, rounded to the nearest integer, is \_\_\_\_\_ g.

- A) 34
- B) 162
- C) 86
- D) 61

48) The molecular formula of aspartame, the generic name of NutraSweet®, is  $C_{14}H_{18}N_2O_5$ . The molar mass of aspartame, rounded to the nearest integer, is \_\_\_\_\_ g.

- A) 24
- B) 156
- C) 294
- D) 43
- E) 39

# Moles Chem Questions And Answers

**Christopher R. Hren, John T.  
Moore, Peter J. Mikulecky**



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