

## 1. Moles and chemical equations :

- Law of conservation of mass :

Total mass of reactants is equal to the total mass of products.

(equation must be balanced)

\*Note :

When you write the balanced equation, the coefficients of balancing are mole ratios between reactants and products.

Examples :



0.5 mole of Al, How many moles  $\text{Al}_2\text{O}_3$  ?

# Moles Study Guide Answers

**RD Boyd**



## **Moles Study Guide Answers:**

## The Enigmatic Realm of **Moles Study Guide Answers**: Unleashing the Language is Inner Magic

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