

Name: _____ Date: _____ Lab Section: _____

Experiment 2: Molecular Models REPORT SHEET

Pre-lab Study Questions:

- Objectives of the experiment (in your own words):
- What kinds of electrons form the covalent bonds in molecules?
- Is a polar molecule symmetrical or asymmetrical?

FORMULA OF MOLECULE	LEWIS STRUCTURE	3-D SHAPE w/ lone pairs	MOLECULAR GEOMETRY	MOLECULAR POLARITY
example: CH_4	$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H} \end{array}$		tetrahedral	nonpolar
H_2O	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$	tetrahedral	polar
H_2O^+	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$	tetrahedral	nonpolar
C_2H_6				
C_2H_4	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$		trigonal planar	non polar
C_2H_2				
CO_2	$\begin{array}{c} \text{O} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \diagup \quad \diagdown \\ \text{O} \end{array}$		Linear	Non polar
H_2O_2				
CHCl_3				

Molecular Model Laboratory Manual Answers

**Ralph Petrucci, William
Harwood, Geoffrey Herring**



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Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

Molecular Modeling for the Design of Novel Performance Chemicals and Materials Beena Rai, 2012-03-23 Molecular modeling MM tools offer significant benefits in the design of industrial chemical plants and material processing operations While the role of MM in biological fields is well established in most cases MM works as an accessory in novel products materials development rather than a tool for direct innovation As a result MM engineers and practitioners are often seized with the question How do I leverage these tools to develop novel materials or chemicals in my industry Molecular Modeling for the Design of Novel Performance Chemicals and Materials answers this important question via a simple and practical approach to the MM paradigm Using case studies it highlights the importance and usability of MM tools and techniques in various industrial applications The book presents detailed case studies demonstrating diverse applications such as mineral processing pharmaceuticals ceramics energy storage electronic materials paints coatings agrochemicals and personal care The book is divided into themed chapters covering a diverse range of industrial case studies from pharmaceuticals to cement While not going too in depth into fundamental aspects the book covers almost all paradigms of MM and references are provided for further learning The text includes more than 100 color illustrations of molecular models

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Manual Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 A Concise Engineering Chemistry Lab Manual for I/II Semester (I Year Mandatory Course) B.E Students Dr. S. R. Pratap, Dr. S. Z. Mohamed Shamshuddin, *Biological Investigations Lab Manual* Warren Dolphin, David Vleck, Linda Westgate, James Colbert, 2010-01-27 The lead author of eight successful previous editions has brought together a team that combined has well over 60 years experience in offering beginning biology labs to several thousand students each year at Iowa State University Their experience and diverse backgrounds ensure that this extensively revised edition will meet the needs of a new generation of students Designed to be used with all majors level general biology textbooks the included labs are investigative using both discovery and hypothesis based science methods Students experimentally investigate topics observe structure use critical thinking skills to predict and test ideas and engage in hands on learning Students are often asked what evidence do you have that in order to encourage them to think for themselves By emphasizing investigative quantitative and comparative approaches to the topics the authors continually emphasize how the biological sciences are integrative yet unique An instructor s manual available through McGraw Hill Lab Central provides detailed advice based on the authors experience on how to prepare materials for each lab teachings tips and lesson plans and questions that can be used in quizzes and practical exams This manual is an excellent choice for colleges and universities that want their students to experience the breadth of modern biology *Food Analysis Laboratory Manual* S. Suzanne Nielsen, 2017-06-07 This third edition laboratory manual was written to accompany Food Analysis Fifth Edition by the same author New to this third edition of the laboratory manual are four introductory chapters that complement both the textbook chapters and the laboratory exercises The 24 laboratory exercises in the manual cover 21 of the 35 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic Most of the laboratory exercises include the following background reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis **Chemistry** Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives **Laboratory Manual of Physical Chemistry** Albert Watson Davison, Henry Sjoerd van Klooster, 1922 *Exploring Anatomy & Physiology in the Laboratory Core Concepts, 2e* Erin C Amerman, 2018-02-01 This brief version of Exploring Anatomy and Physiology in the Laboratory 3e is intended for one semester anatomy and physiology courses geared toward allied health students Exploring Anatomy Physiology Laboratory Core Concepts by Erin C Amerman is a comprehensive beautifully illustrated and affordably priced lab manual that features an innovative interactive approach to engage your students and help ensure a deeper understanding of

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