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#### CHAPTER 11 REVIEW

# Molecular Composition of Gases

#### MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided.

- 1. \_\_\_\_\_ The average speed of a gas molecule is most directly related to the \_\_\_\_
  - (a) polarity of the molecule
  - (b) pressure of the gas
  - (c) temperature of the gas
  - (d) number of moles in the sample
- Which of the following statements best represents Graham's law for two gases at the same temperature?
  - (a) The molar masses and the velocities are inversely proportional.
  - (b) The molar masses and the velocities are directly proportional.
  - (c) The molar masses and the square roots of the velocities are directly proportional.
  - (d) The molar masses and the squares of the velocities are inversely proportional.
- 3. Water is synthesized from its elements according to the following equation:

$$2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$$

- a. Which is greater, the volume of H<sub>2</sub> consumed or the volume of O<sub>2</sub> consumed?
- b. Which is greater, the mass of H<sub>2</sub> consumed or the mass of O<sub>2</sub> consumed?
- 4. Consider a 2.24 L sample of an ideal gas at STP.
  - \_\_\_\_\_ a. How many moles are present in the sample, or is more data needed?
  - b. How many molecules are present in the sample, or is more data needed?
  - e. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

# **Molecular Compositions Of Gases Answers**

**M** Carnoy

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transfer processes Since they tend to have the greatest effect on the reliability of processes they power compressors typically receive the most scrutiny of all the machinery among the general population of processing equipment To prevent unwanted compressor failures from occurring operators must be taught how their equipment should operate and how each installation is different from one another The ultimate purpose of this book is to teach those who work in process settings more about gas compressors so they can start up and operate them correctly and monitor their condition with more confidence Some may regard compressor technology as too broad and complex a topic for operating personnel to fully understand but the author has distilled this vast body of knowledge into some key easy to understand lessons for the reader to study at his or her own pace This groundbreaking new work is a must have for any engineer operator or manager working with process compressors The main goals of this book are to Explain important theories and concepts about gases and compression processes with a minimum of mathematics Identify key compressor components and explain how they affect reliability Explain how centrifugal compressors reciprocating compressors and screw compressors function Explain key operating factors that affect reliabilityIntroduce the reader to basic troubleshooting methodologies Introduce operators to proven field inspection techniques Improve the confidence of personnel operating compressors by teaching them the basics of compressor theory Improve compressor reliability plantwide by teaching operating and inspection best practices Improve communication between operating and supporting plant personnel by providing a common vocabulary of compressor terms Help processing plants avoid costly failures by teaching operators how to identify early compressor issues during field inspections

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orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry

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