

Name: _____ **KEY** _____ Period: _____ Date: _____

Determining Molecular Polarity

Molecules are either POLAR or NON-POLAR. This is determined by:

- If all of the bonds are ionic → the compound is just "ionic"
- If all of the bonds are non-polar covalent → then the molecule is non-polar
- If one of the bonds is polar → then the molecule is polar
- If there is more than one polar bond, then look at symmetry
 - If the shape linear, tetrahedron, or trigonal planar and all the bonds are the same, then it had symmetry and is non-polar.
 - If the shape is a pyramid, linear, or bent, then the molecules is NOT symmetrical and is a polar molecule

NPC = Non-Polar Covalent
 PC = Polar Covalent
 I = Ionic

Fill in the following table:

Compound	Lewis Dot Diagram	Bond Type (NPC/PC/I)	Shape	Symmetry (Y/N)	Molecular Polarity (P/NP)
H ₂	H — H	H—H (NPC)	Linear	YES	NP
N ₂	:N ≡ N:	N—N (NPC)	Linear	YES	NP
O ₂	:Ö = Ö:	O—O (NPC)	Linear	YES	NP
CaCl ₂	:Cl: Ca :Cl:	(I)	No Shape (Ionic)	NONE	IONIC
CO	:C ≡ O:	C—O (PC)	Linear	NO	P

Molecular Shape And Polarity Study Guide Answers

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**



Molecular Shape And Polarity Study Guide Answers:

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Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph

Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications *Chemical Principles Student's*

Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text **Basic Concepts of**

Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter **Organic Chemistry Study Guide** Robert J. Ouellette, J. David

Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic

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engaging *Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston, Denniston Katherine, 2006 *General Chemistry Workbook* Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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