

Name: \_\_\_\_\_ **KEY** \_\_\_\_\_ Period: \_\_\_\_\_ Date: \_\_\_\_\_

## Determining Molecular Polarity

Molecules are either POLAR or NON-POLAR. This is determined by:

- If all of the bonds are ionic → the compound is just "ionic"
- If all of the bonds are non-polar covalent → then the molecule is non-polar
- If one of the bonds is polar → then the molecule is polar
- If there is more than one polar bond, then look at symmetry
  - If the shape linear, tetrahedron, or trigonal planar and all the bonds are the same, then it had symmetry and is non-polar.
  - If the shape is a pyramid, linear, or bent, then the molecules is NOT symmetrical and is a polar molecule

NPC = Non-Polar Covalent  
 PC = Polar Covalent  
 I = Ionic

Fill in the following table:

Compound	Lewis Dot Diagram	Bond Type (NPC/PC/I)	Shape	Symmetry (Y/N)	Molecular Polarity (P/NP)
H <sub>2</sub>	H — H	H—H (NPC)	Linear	YES	NP
N <sub>2</sub>	:N ≡ N:	N—N (NPC)	Linear	YES	NP
O <sub>2</sub>	:Ö = Ö:	O—O (NPC)	Linear	YES	NP
CaCl <sub>2</sub>	:Cl: Ca :Cl:	(I)	No Shape (Ionic)	NONE	IONIC
CO	:C ≡ O:	C—O (PC)	Linear	NO	P

# **Molecular Shape And Polarity Study Guide Answers**

**Denise Guinn, Rebecca Brewer, Rachel  
C. Lum**



## **Molecular Shape And Polarity Study Guide Answers:**

**Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

**Chemical Principles Study Guide/Solutions Manual** John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

**Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

Study Guide/Solutions Manual for Organic Chemistry K. Peter C. Vollhardt, Neil E. Schore, 2018-01-12 Updated for the Eighth Edition of Vollhardt Schore Organic Chemistry and written by the book s coauthor Neil Schore this invaluable manual includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us **Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1992

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