

3. Refer to the data table in Question 2.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction shown in Model 1?

The ratios are the same as the coefficients in Model 1: 1:3:2.

b. Use mathematical concepts to explain how your answer in part a is possible.

The coefficients were all multiplied by the same number, so they all ended back to the same thing.

4. Even 538 is a small number of molecules to use in a reaction. Typically chemists use much larger numbers of molecules. (Recall that one mole is equal to 6.02×10^{23} particles.) Consider each situation below as it relates to the reaction in Model 1: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole number possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$
If the reaction occurred 6.02×10^{23} times, how many molecules would be consumed or produced?	6.02×10^{23}	1.81×10^{24}	1.20×10^{24}	1 : 3 : 2
How many <i>moles</i> of each substance would be consumed or produced in the previous situation?	1	3	2	1 : 3 : 2

5. Refer to the data table in Question 4.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction in Model 1?

The reduced ratios are the same as the coefficients.

b. Use mathematical concepts to explain how your answer in part a is possible.

We are still multiplying coefficients by the same number so we just didn't everything by the number to get back to a reduced ratio.

6. The ratio obtained from the coefficients in a balanced chemical equation is called the **mole ratio**.

a. What is the mole ratio for the reaction in Model 1?



b. Explain why this ratio is called the mole ratio?

It is the reduced ratio in moles (all the coefficients = 6.02×10^{23}).

Mole Ratios Answer Key Pogil

Brendan G. Carr

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Mole Ratios Answer Key Pogil

1. Understanding the eBook Mole Ratios Answer Key Pogil
 - The Rise of Digital Reading Mole Ratios Answer Key Pogil
 - Advantages of eBooks Over Traditional Books
2. Identifying Mole Ratios Answer Key Pogil
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mole Ratios Answer Key Pogil
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mole Ratios Answer Key Pogil
 - Personalized Recommendations
 - Mole Ratios Answer Key Pogil User Reviews and Ratings
 - Mole Ratios Answer Key Pogil and Bestseller Lists
5. Accessing Mole Ratios Answer Key Pogil Free and Paid eBooks
 - Mole Ratios Answer Key Pogil Public Domain eBooks
 - Mole Ratios Answer Key Pogil eBook Subscription Services
 - Mole Ratios Answer Key Pogil Budget-Friendly Options

6. Navigating Mole Ratios Answer Key Pogil eBook Formats
 - ePub, PDF, MOBI, and More
 - Mole Ratios Answer Key Pogil Compatibility with Devices
 - Mole Ratios Answer Key Pogil Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mole Ratios Answer Key Pogil
 - Highlighting and Note-Taking Mole Ratios Answer Key Pogil
 - Interactive Elements Mole Ratios Answer Key Pogil
8. Staying Engaged with Mole Ratios Answer Key Pogil
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mole Ratios Answer Key Pogil
9. Balancing eBooks and Physical Books Mole Ratios Answer Key Pogil
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mole Ratios Answer Key Pogil
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mole Ratios Answer Key Pogil
 - Setting Reading Goals Mole Ratios Answer Key Pogil
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mole Ratios Answer Key Pogil
 - Fact-Checking eBook Content of Mole Ratios Answer Key Pogil
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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