

Dilution, Molarity, and Solutions

Name _____	Section _____	Date _____
Molarity and Dilution		
Directions:		
1) 33.6 L of a 6.00 M solution of NaCl are diluted to 33.9 L. What is the new molarity of the solution?		
2) An aqueous solution of H_2SO_4 with an original molarity of 1.35M is diluted to a volume of 2.34 L and a molarity of 0.835M. What was its original volume?		
3) A solution of H_3PO_4 in water is concentrated to a volume of 0.49 L and a molarity of 2.23 M. What was its original molarity if its original volume was 0.95 L?		
4) What is the new volume of a solution of H_2CrO_4 with a molarity of 3.14M and a volume of 2.87L if its molarity is changes to 0.698 L?		
5) To make a 3.99M solution of HCl from 338 mL of a 6.02M solution, what volume must the new solution be in liters?		

$$M_1V_1 = M_2V_2$$

moles before = moles after

A CHEMISTRY WORKSHEET

10 PROBLEMS

WORKED ANSWER KEY INCLUDED

Molarity By Dilution Page 69 Answers

Daniela Niemeyer

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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