Chemistry: Molarity

Directions: Using these formulas, solve the following problems



molarity = mol solute / L solution 1 amu = 1 gram / mol

M.W. = MIW.

Problem 1: Calculate the molarity of a 2.21 L solution made with 317 g of calcium chloride Answer Problem 2. How many grams of NaCl would be required to produce a 6.1 M solution with a volume of 0.64 L? Answer. Problem 3. How many grams of NaCH would be required to produce a 5.36 M solution. with a volume of 1.37 L7 Answer. Problem 4. How many grams of KCI would you need to weigh out to prepare 0.416 L of a 3.37 M KCI salution? Answer. Problem 5: Calculate the molarity of a 6.68 mt, solution of 6.89 a NaCl. Answer ____ Problem 6: Calculate the volume in ml. if you have 18.9 g of Na₂CO₂ from a 0.112 M solution Answer Problem 7. Calculate the volume in mL if you have 28.1 g LiOH from a 7.11 M solution. Arthwest Problem B. Calculate the final concentration if water is added to 0.789 L of a 3.68 M HCI solution to give a volume of 0.387 L. Answer Problem 9: Calculate the final concentration if a 8.63 mL of a 1.12 M HCI solution is diluted with water to 2.54 mL. Answer. Problem 10: Caliculate the final concentration if water is added to 0.235 L of a 6.11 M KBr solution to give a volume of 5.79 L. Answer.

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Ying-Ying Zheng

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Reviewing Molarity Worksheet Answers Page 273: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "Molarity Worksheet Answers Page 273," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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