

Mole Calculations Review

Molar Mass Practice

Find the molar masses of the following compounds:

- | | | | |
|---|--|--|--------------------------------------|
| 1) NaBr | 2) PbSO ₄ | 3) Ca(OH) ₂ | 4) Na ₃ PO ₄ |
| 5) (NH ₄) ₂ CO ₃ | 6) C ₆ H ₁₂ O ₆ | 7) Fe ₂ (PO ₄) ₂ | 8) (NH ₄) ₂ S |
| 9) Zn(C ₂ H ₃ O ₂) ₂ | 10) AgF | | |

Answers (rounded to the nearest 0.1 grams)

- 1) 102.9 g/mol, 2) 303.3 g/mol, 3) 74.1 g/mol, 4) 164.0 g/mol
5) 96.0 g/mol, 6) 180.0 g/mol, 7) 357.4 g/mol, 8) 68.1 g/mol
9) 183.4 g/mol, 10) 126.9 g/mol

Mass - Moles - Particles Practice

- How many moles of Na are in 42 g of Na?
- How many moles of O are in 8.25 g of O?
- What is the mass of 2.18 mol of Cu?
- What is the mass of 0.28 mol of iron?
- How many atoms are in 7.2 mol of chlorine?
- How many atoms are in 36 g of bromine?
- How many moles are in 1.0×10^9 atoms?
- What is the mass of 1.20×10^{25} atoms of sulfur?
- How many moles of CO molecules are in 52 g of CO?
- How many moles of C₂H₆ are in 124 g?
- How many moles of CCl₄ are there in 56 g of CCl₄?
- What is the mass of 2.50 mol of H₂SO₄?
- What is the mass of 0.25 mol of Fe₂O₃?
- How many molecules are there in 52 g of CO?
- How many formula units are in 22.4 g SnO₂?
- How many molecules are in 116 g CCl₄?
- What is the mass of 3.01×10^{23} formula units of Fe₂O₃?
- What is the mass of 1.2×10^{25} molecules of CO?
- How many O atoms are in 1.25 mol of SO₂?
- How many moles of O atoms are there in 1.20×10^{25} N₂O₅ molecules?
- How many formula units are in 5.33 mol of CuCl₂?
- How many copper atoms are in 5.33 mol of CuCl₂?
- How many moles of Cl atoms are in 5.33 mol of CuCl₂?

Molar Mass Practice Answers

Judd E. Hollander

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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