Mixed Stoichiometry Problems

- How many moles of H₂ would be required to completely react with O₂ to produce 5 moles of water? 5 mol H₂
- H₂SO₄ + NaOH → Na₂SO₄ + H₂O
 - a. Balance this equation
 - b. What mass of H₂SO₄ would be required to react with 0.75 mol of NaOH? 37g
- What mass of NO₂ is formed when NO reacts with 384 g of O₂? 1104g
- If 30 g of H₂SO₄ reacts with aluminum hydroxide in a double replacement reaction, what mass of water is produced? 11.0g
- If 10 L of CH₄ combusts with O₂, what volume of water will be produced? 20 L H₂O
- Tin(II) fluoride, SnF₂, is used in some toothpastes. It is made by the reaction of tin with hydrogen fluoride according to the following equation.

 $Sn + 2HF \rightarrow SnF_2 + H_2$

How many grams of SnF_2 are produced from the reaction of 30 g of HF with Sn? 118 g

In a spacecraft, the carbon dioxide exhaled by astronauts can be removed by its reaction with lithium hydroxide, LiOH, according to the following chemical equation.

 $CO_2(g) + 2LiOH(s) \rightarrow Li_2CO_3(s) + H_2O(l)$

If 448 L of CO₂ is exhaled, the average amount exhaled by a person each day, how much (in grams) Li₂CO₃(s) is produced? 1480g Li₂CO₃

- When 9.8g of aluminum oxide decomposes, how many grams of aluminum metal are produced? 5.2 g Al
- How many grams of iodine are produced when 0.72 mol of fluorine react with potassium iodide?
- How many grams of sodium are required to react with water to produce 5.0g of sodium hydroxide? (Na + H₂O → NaOH + H₂)
- 11. How many grams of H2 are produced in the reaction in #10?
- 12. What volume would be occupied by the resulting hydrogen at STP?
- 13. 20.0g of K react with H₂O to produce KOH and H₂ gas. How many liters would the hydrogen gas occupy at STP?
- 14. If 30.2g of Al react with HCl to produce AlCl₃ and H₂ gas, how many liters of H₂ are produced at STP?
- 15. How many liters of HCl are produced by the reaction of 5.7L of H₂ with Cl₂?

Modern Chemistry Stoichiometry Mixed Review Answers

G Thomas

Modern Chemistry Stoichiometry Mixed Review Answers:

The Top Books of the Year Modern Chemistry Stoichiometry Mixed Review Answers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have captivated audiences this year. The Must-Read: Colleen Hoovers "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best: Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic: Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a guiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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