

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme sollen Sie zur Vertiefung des wissenschaftlichen Verständnisses der Stoffeigenschaften (Kapitel 10) befähigen. Sie sind, insgesamt für die vollständige Bearbeitung, 120 Minuten rechnerischer Arbeit zu leisten.

$$M(\text{H}_2\text{O}) = 18,015 \text{ g/mol}$$

1. Wasser siedet bei Normaldruck (1013 hPa) bei 100 °C. Welchen Anteil an Wasser verdunstet, wenn man 100 g Wasser bei 100 °C erwärmt?

$$m = \frac{100 \text{ g} \cdot (1013 \text{ hPa} - 1013 \text{ hPa})}{1013 \text{ hPa}} = 0 \text{ mol}$$

2. Ethanol siedet bei Normaldruck (1013 hPa) bei 78 °C. Welchen Anteil an Ethanol verdunstet, wenn man 100 g Ethanol bei 78 °C erwärmt?

$$m = \frac{100 \text{ g} \cdot (1013 \text{ hPa} - 1013 \text{ hPa})}{1013 \text{ hPa}} = 0 \text{ mol}$$

3. Ethanol ist ein Gemisch aus Ethanol und Wasser. Die Zusammensetzung ist 100 g Ethanol in 100 g Wasser. Wie hoch ist der Dampfdruck bei 25 °C?

$$p = \frac{100 \text{ g} \cdot 1013 \text{ hPa} + 100 \text{ g} \cdot 1013 \text{ hPa}}{200 \text{ g}} = 1013 \text{ hPa}$$

4. In der Natur findet man oft Gemische aus Ethanol und Wasser. Die Zusammensetzung ist 100 g Ethanol in 100 g Wasser. Wie hoch ist der Dampfdruck bei 25 °C? (Die Dampfdruckkurven für Ethanol und Wasser sind in der Tabelle angegeben.)

$$p = \frac{100 \text{ g} \cdot 1013 \text{ hPa} + 100 \text{ g} \cdot 1013 \text{ hPa}}{200 \text{ g}} = 1013 \text{ hPa}$$

5. Ethanol ist ein Gemisch aus Ethanol und Wasser. Die Zusammensetzung ist 100 g Ethanol in 100 g Wasser. Wie hoch ist der Dampfdruck bei 25 °C? (Die Dampfdruckkurven für Ethanol und Wasser sind in der Tabelle angegeben.)

$$p = \frac{100 \text{ g} \cdot 1013 \text{ hPa} + 100 \text{ g} \cdot 1013 \text{ hPa}}{200 \text{ g}} = 1013 \text{ hPa}$$

6. Ethanol ist ein Gemisch aus Ethanol und Wasser. Die Zusammensetzung ist 100 g Ethanol in 100 g Wasser. Wie hoch ist der Dampfdruck bei 25 °C? (Die Dampfdruckkurven für Ethanol und Wasser sind in der Tabelle angegeben.)

Modern Chemistry Section 3 Gases Answer Key

M Mosston



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