Ch. 9 Practice Test

Chapter: Stoichiometry

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- 1. A balanced chemical equation allows one to determine the
 - a. mole ratio of any two substances in the reaction.
 - b. energy released in the reaction.
 - c. electron configuration of all elements in the reaction.
 - mechanism involved in the reaction.
- 2. The coefficients in a chemical equation represent the
 - a. masses, in grams, of all reactants and products.
 - relative numbers of moles of reactants and products.
 - c. number of atoms in each compound in a reaction.
 - d. number of valence electrons involved in the reaction.
- 3. To determine the limiting reactant in a chemical reaction involving known masses of the two reactants, which of the following would be most useful?
 - a. determining the masses of 100 mol of each reactant
 - b. determining the molar masses of the products
 - c. calculating bond energies
 - d. calculating the mass of a single product formed from each reactant
- 4. How many mole ratios can be correctly obtained from the chemical equation 2Al₂O₃(I) → 4Al(x) + 3O₂(g)
 - a. 3
 - b. 4
 - c. 6
 - d. 8.

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Mark Steyvers

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