

Name _____ Class _____ Date _____

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H_2SO_4
 - H_3PO_4
 - HNO_3
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH^- ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO_3^-
 - H_3O^+
 - H_3PO_4
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(aq) + \text{CH}_3\text{NH}_2(aq) \rightleftharpoons \text{HC}_2\text{O}_4^-(aq) + \text{CH}_3\text{NH}_3^+(aq),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(aq)$ and $\text{CH}_3\text{NH}_2(aq)$
 - $\text{H}_2\text{C}_2\text{O}_4(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$
 - $\text{CH}_3\text{NH}_2(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$
 - $\text{HC}_2\text{O}_4^-(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$

Modern Chemistry Chapter 21 Test

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Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2023 Organic Chemistry 13th edition provides a comprehensive yet accessible treatment of all the essential organic chemistry concepts with emphasis on relationship between structure and reactivity in the subject The textbook includes all the concepts covered in a typical organic chemistry textbook but is unique in its skill development approach to the subject Numerous hands on activities and real world examples are integrated throughout the text to help students understand both the why and the how behind organic chemistry This International Adaptation offers new and updated content with improved presentation of all course material It offers new material on several topics including the relevance of intermolecular forces in the immune response and vaccines like those for Covid 19 the chemistry of breathing carbonic anhydrase how conjugation and complexation affect the color of lobsters and how biodegradable polymers are used to stabilize vaccines and pharmaceuticals Content is revised to reflect the current understanding of chemical processes and improved depictions of longstanding mechanisms This edition builds on the ongoing pedagogical strength of the book with the inclusion of additional worked and end of chapter problems and an engaging set of new problems entitled Chemical Consultant Needed These draw from the primary chemical literature and give students experience of working with more complex polyfunctional structures and areas where key transformations take place *Principles of Modern Chemistry* David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom Modern Microcrystal Tests for Drugs Charles Clarke Fulton, 1969 **Lab Experiments Modern Chemistry** Harold Clark Metcalfe, 1982 **Laboratory Experiments to Accompany "Modern Chemistry,"** Fredus Nelson Peters, 1902 Biomedical Aspects of the Laser Leon Goldman, 2013-11-09 This book is a review of past and current studies and future plans of the Laser Laboratory in Cincinnati and some of the contributions of laser research groups in other medical centers Special thanks are due to the Directing Physicist of the Laser Laboratory R James Rockwell Without his advice constant supervision and corrections this enthusiastic investigator would continue to upset even many more people than he has done already The excuse of course is to stimulate much needed interest and controlled research and

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Chemical Weapons

Convention Chemicals Analysis Markku Mesilaakso, 2005-12-13 Describes the procedures for collection of samples sample preparation and analysis of CWC related chemicals It deals with analytical procedures that can be followed in well equipped off site laboratories designated laboratories as well as the on site analytical procedures that the OPCW inspectors use in sample collection and preliminary analysis of the samples in field conditions A one of a kind highly topical handbook for every expert in the chemical weapons field Outlines the methods for analysing chemical weapons both on and off site Authored by international experts in the field from top laboratories in both government and academic institutions

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