

- (i) It depends on the concentration of a non-electrolyte solute in a solution and does not depend on the nature of the solute molecules.
- (ii) It depends on a number of particles of electrolyte solute in a solution and does not depend on the nature of the solute particles.
- (iii) It depends on the concentration of a non-electrolyte solute in solution as well as on the nature of the solute molecules.
- (iv) It depends on the concentration of an electrolyte or nonelectrolyte solute in solution as well as on the nature of solute molecules.

Solution:

Option (i) and (ii) are the answers.

30. Van't Hoff factor i is given by the expression _____.

- (i) $i = \text{Normal molar mass} / \text{Abnormal molar mass}$
- (ii) $i = \text{Abnormal molar mass} / \text{Normal molar mass}$
- (iii) $i = \text{Observed colligative property} / \text{calculated colligative property}$
- (iv) $i = \text{Calculated colligative property} / \text{Observed colligative property}$

Solution:

Option (i) and (iii) are the answers.

31. Isotonic solutions must have the same _____.

- (i) solute
- (ii) density
- (iii) elevation in boiling point
- (iv) depression in freezing point

Solution:

Option (iii) and (iv) are the answers.

32. Which of the following binary mixtures will have the same composition in the liquid and vapour phase?

- (i) Benzene - Toluene
- (ii) Water-Nitric acid
- (iii) Water-Ethanol
- (iv) n -Hexane - n -Heptane

Solution:

Option (ii) and (iii) are the answers.

33. In isotonic solutions _____.

- (i) solute and solvent both are same.
- (ii) osmotic pressure is the same.
- (iii) solute and solvent may or may not be same.
- (iv) a solute is always the same solvent may be different.

Solution:

Option (ii) and (iii) are the answers.

34. For a binary ideal liquid solution, the variation in total vapour pressure versus the composition of the solution is given by which of the curves?

Modern Chemistry Chapter 12 Section 2 Review Answers

Ian Pickup



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Lab Experiments Modern Chemistry* Harold Clark Metcalfe, 1982 *Fundamentals of Ecotoxicology* Michael C. Newman, 2019-11-27 This new edition is revised throughout and includes new and expanded information on natural resource damage assessment the latest emerging contaminants and issues and adds new international coverage including case studies and rules and regulations The text details key environmental contaminants explores their fates in the biosphere and discusses bioaccumulation and the effects of contaminants at increasing levels of ecological organization Vignettes written by experts illustrate key themes or highlight especially pertinent examples This edition offers an instructors solution manual PowerPoint slides and supplemental images Features Adds all new discussions of natural resource damage assessment concepts and approaches Includes new vignettes written by leading guest authors Draws on materials from 2 500 cited sources including 400 new to this edition Adds numerous new entries to a useful glossary of 800 terms Includes a new appendix discussing Brazilian environmental laws and regulations added to existing appendices outlining U S E U Chinese Australian and Indian environmental laws *Fundamentals of Ecotoxicology* The Science of Pollution Fifth Edition contains a broad overview of ecotoxicology and provides a basic understanding of the field Designed as a textbook for use in introductory graduate or upper level undergraduate courses in ecotoxicology applied ecology environmental pollution and environmental science it can also be used as a general reference for practicing environmental toxicologists *Biochemistry* John T. Tansey, 2019-04-02 *Biochemistry An Integrative Approach* is addressed to premed biochemistry and life science majors taking a one semester biochemistry course This version includes the first 12 chapters and should only be used for one semester biochemistry courses *Biochemistry* addresses the diverse needs of premed biochemistry and life science majors by presenting relevant material while still preserving a chemical perspective Presented within the next generation of WileyPLUS *Biochemistry* emphasizes worked problems through video walkthroughs interactive elements and expanded end of chapter problems with a wide range of subject matter and difficulty The worked problems in the course are both qualitative and quantitative and model for students the biochemical reasoning they need to practice Students will often be asked to analyze data and make critical assessments of experiments *Principles of Modern Chemistry* David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses *PRINCIPLES OF MODERN CHEMISTRY* continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most

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and the other showing that the position of maximum disorder of the water molecules occurs at a charge opposite to that needed for interpretations of capacitance humps in terms of water molecules M A Habib who has contributed to the theory in this area reviews the consequences of these changes in information The rise in the price of energy toward a situation in which sources other than the fossil fuels become economical implies much for the fuel cell and electrocatalysis It has long been known that electrocatalysis in real situations was more than a consideration of exchange current densities and a gap remains in the formulation of the theory of supports for such catalysts although Boudart has stressed so much the vital nature of them P Stonehart and K A Kinoshita describe progress in this area *Modern Nanotechnology* Junaid Ahmad Malik, Mohamed Jaffer Sadiq Mohamed, 2023-07-18 This two volume set provides a comprehensive overview of modern nanoscience and encompasses advanced techniques of nanocomposite materials that make their way from the laboratory to the field for the revival of energy and environmental systems in a sustainable manner It includes the design and the sophisticated fabrication of nanomaterials along with their potential energy and environmental applications while looking at how nanoscience and nanotechnology can be used to promote environmentally friendly processes and strategies The books purpose is to promote eco friendly methods and techniques by covering many elements of both the synthesis and uses of nanoparticles and nanofluids for energy and environmental engineering They provide an up to date synthesis of nanocomposite materials for modern nanotechnology applications in the fields of environment protection heterogeneous catalysis wastewater treatment fuel cells electrochemical energy conversion and storage applications The set is designed for environmental scientists nanotechnologists chemists engineers and individuals seeking current research on nanotechnology and its applications in environmental engineering Graduate students working in these fields will also find it a valuable resource Volume 1 focuses on the fundamentals of nanotechnology environmental protection sustainable agriculture bioremediation bio nanocomposites and wastewater treatment Biochemistry, International Adaptation Donald Voet, Judith G. Voet, 2021 The Gold Standard in Biochemistry text books Biochemistry 4e is a modern classic that has been thoroughly revised Don and Judy Voet explain biochemical concepts while offering a unified presentation of life and its variation through evolution It incorporates both classical and current research to illustrate the historical source of much of our biochemical knowledge **Pharmaceutical Journal** ,1961 Literary Gazette and Journal of Belles Lettres, Arts, Sciences, &c ,1851 *Encyclopedia of Electrochemical Power Sources* ,2024-10-03 The Encyclopedia of Electrochemical Power Sources Second Edition is a comprehensive seven volume set that serves as a vital interdisciplinary reference for those working with batteries fuel cells electrolyzers supercapacitors and photo electrochemical cells With an increased focus on the environmental and economic impacts of electrochemical power sources this work not only consolidates extensive coverage of the field but also serves as a gateway to the latest literature for professionals and students alike The field of electrochemical power sources has experienced significant growth and development since the first edition was published in 2009 This is reflected in the

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