

CHAPTER 1 REVIEW***Matter and Change*****SECTION 2**

SHORT ANSWER Answer the following questions in the space provided.

- 1.** Classify each of the following as a *homogeneous* or *heterogeneous* substance.

<u>heterogeneous</u>	a. iron ore
<u>homogeneous</u>	b. quartz
<u>heterogeneous</u>	c. granite
<u>homogeneous</u>	d. energy drink
<u>heterogeneous</u>	e. oil-and-vinegar salad dressing
<u>homogeneous</u>	f. salt
<u>homogeneous</u>	g. rainwater
<u>homogeneous</u>	h. nitrogen

- 2.** Classify each of the following as a *physical* or *chemical* change.

<u>physical</u>	a. ice melting
<u>chemical</u>	b. paper burning
<u>chemical</u>	c. metal rusting
<u>physical</u>	d. gas pressure increasing
<u>physical</u>	e. liquid evaporating
<u>chemical</u>	f. food digesting

- 3.** Compare a physical change with a chemical change.

A chemical change involves a rearrangement of the atoms of different elements in a
substance and the formation substances with different physical properties. A physical
change can occur in properties such as the state or shape of a substance, but it will
not affect the composition of that substance.

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