

Question Number	Acceptable Answers	Reject	Mark
21(a)(v)	Correct final answer (181/182) to 3SF with or without working scores (2) Answer to (iv) x 45 OR x 56.1 (1) $n = 7.18 \times 10^{-2} \times 45 = (3.231) \text{ (mol)}$ $m = 3.231 \times 56.1 = (181.359 / 181.4)$ OR $181.2591 / 181.3$ $= 181 \text{ (g)}$ NOTE ALLOW USE OF 56 (1) Alternative method Answer to (ii) x $\frac{45000}{25}$ OR x 56/56.1 (1) Amount = 0.81 (mol) Mass of KOH left $0.81 \times 56.1/56$ $= 45.441/45.36 \text{ (g)}$ Mass used = $226.8 - 45.441/45.36$ $= 181 \text{ (g)}$ (1)		2

Some TE values:

Part	Answer	Mark	Answer	Mark
(i)	$\frac{25 \times 0.05}{1000}$ $= 1.25 \times 10^{-3}$	0	$\frac{25 \times 4.5}{1000}$ $= 0.1125$	0
(ii)	2.5×10^{-3}	1	0.225	1
(iii)	0.1	1	9	1
(iv)	$0.089 - 0.1$ $= -0.0102$	2	$0.089 - 9$ $= -8.91$	2
(v)	0.459 (mol) and 25.7 (g)	2	400.95 (mol) and 22 500 (g)	2

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