

**PUNJAB PUBLIC SERVICE COMMISSION  
COMBINED COMPETITIVE EXAMINATION  
FOR RECRUITMENT TO THE POSTS OF  
PROVINCIAL MANAGEMENT SERVICE, ETC.**

**CHEMISTRY (OPTIONAL) PAPER-I**

**TIME ALLOWED: 3 HOURS**

**MAXIMUM MARKS: 100**

**Note:** Attempt any FIVE questions only. Write answers in English.

- Q.1:** (a) Using a diagram, explain the variation of molecular energy with internuclear separation. (10)
- (b) What is the primary assumption of the VSEPR model? Explain the difference between square-pyramidal and trigonal bi-pyramidal arrangement in terms of VSEPR. Predict the shape of an  $SeF_4$  molecule. (10)
- Q.2:** (a) What are the mineral sources for copper? Describe in detail the information which can be obtained for 3d metals using a FROST diagram. (10)
- (b) (i) On the basis of trends in the 3d series, suggest possible  $M^{2+}$  aqua ions for use as reducing agents, and write a balanced equation for the reaction of one of these ions with  $O_2$  in acidic solution. (5)
- (ii) Comment on the 2+ oxidation states of 4d and 5d elements. (5)
- Q.3:** (a) Explain with the help of a flow sheet diagram, the electrolytic process of caustic soda manufacturing. (10)
- (b) Give a detailed comparison between dry and wet processing of cement manufacturing. (10)
- Q.4:** (a) Discuss the physical origin of quantization energy for a particle confined to moving inside a one dimensional box or on a ring. (10)
- (b) Use EHT to calculate the energies of the three molecular orbitals in an  $H_3^+$  molecule ion treated as an equilateral triangle of side  $R = 3.25a_0$  (170 pm). (10)
- Q.5:** (a) Calculate the work done when 50g of iron reacts with HCl to produce hydrogen gas in: (5)
- (i) a closed vessel of fixed volume, (ii) an open beaker at  $25^\circ C$ .
- (b) The molar constant pressure heat capacity of a certain solid at 10K is  $0.43 J K^{-1} mol^{-1}$ . What is its molar entropy at that temperature? (10)
- Q.6:** (a) Suggest a calibration procedure for a  $Na^+$  selective electrode, in which the values of  $E_{ap}$ ,  $\beta$  and  $k_{xy}$  are measured experimentally. Assume that  $K^+$  is the only interfering species. (10)
- (b) Distinguish between galvanic and electrolytic cells. (10)
- Q.7:** (a) Define Nuclear Fission and Nuclear Fusion processes. Discuss their comparative features. (10)
- (b) Outline a method for preparation of CAN pellets. Distinguish between calcium superphosphate and Triple Superphosphate. (10)

# June 2012 66ch01 Chemistry Past Paper

**RD Boyd**



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
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