

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

Periodic Table

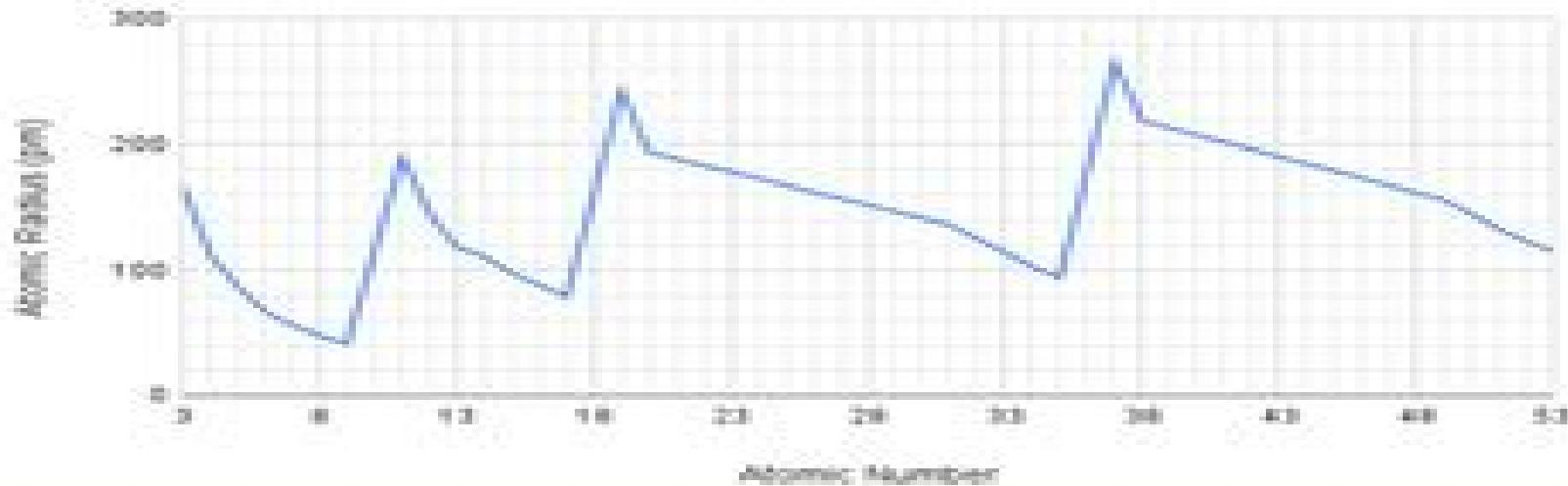
Graphing Periodic Table Trends

DIRECTIONS: Using the provided graph paper, sketch each of the elemental properties using a line graph. Place the element's atomic number on the x-axis and the atomic property on the y-axis. Sketch as clean, neaten vertical lines on each graph where each atom sits on the periodic table grid. (i.e. atomic number 10) Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	1157	Boron - 71	1191	Fluorine - 79	1243	Magnesium - 77	1265
Neon - 4	1112	Magnesium - 12	1175	Sodium - 201	1304	Silicon - 269	1319
Fluorine - 5	1077	Aluminum - 13	1198	Chlorine - 311	1335	Iron - 459	1346
Carbon - 6	1037	Boron - 14	1111	Sulfur - 327	1375	Tin - 503	1389
Hydrogen - 7	995	Phosphorus - 15	1081	Argon - 362	1414	Antimony - 511	1429
Silicon - 8	955	Boron - 16	1046	Boron - 364	1453	Tellurium - 532	1458
Fluorine - 9	927	Chlorine - 17	1178	Fluorine - 369	1484	Yttrium - 563	1495

Periodic Table Trends (continued)							
Lithium - 3	1157	Boron - 71	1191	Fluorine - 79	1243	Magnesium - 77	1265
Neon - 4	1112	Magnesium - 12	1175	Sodium - 201	1304	Silicon - 269	1319
Fluorine - 5	1077	Aluminum - 13	1198	Chlorine - 311	1335	Iron - 459	1346

Atomic Radius



Graphing Trends In The Periodic Table Answers

Randy L. Bell,Joe Garofalo

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communicate health findings Each chapter features 10 20 mastery check questions with detailed answers 5 optional problems A case study A multiple choice short answer quiz Answers to the cases and quizzes are provided as part of the online instructor resource package [Understanding Advanced Physical Inorganic Chemistry: The Learner's Approach \(Revised Edition\)](#) Kim Seng Chan,Jeanne Tan,2016-09-26 This revised edition has been updated to meet the minimum requirements of the new Singapore GCE A level syllabus that would be implemented in the year 2016 Nevertheless this book is also highly relevant to students who are studying chemistry for other examination boards In addition the authors have also included more Q A to help students better understand and appreciate the chemical concepts that they are mastering

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