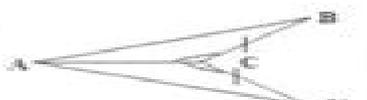
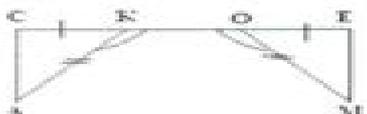
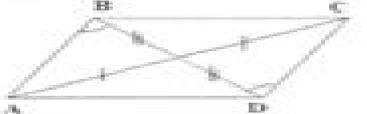
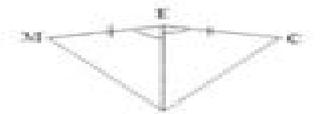
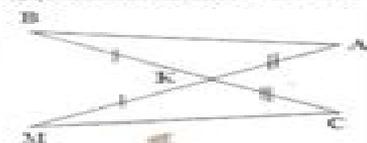
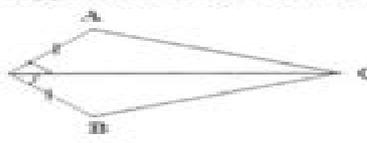
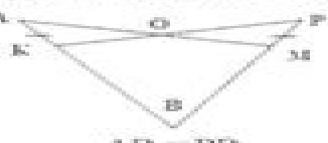
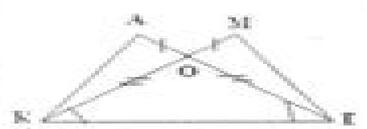
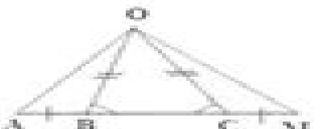
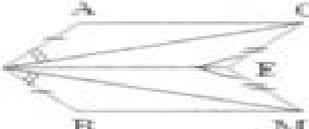
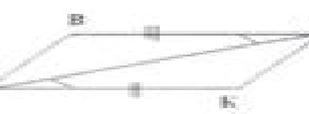
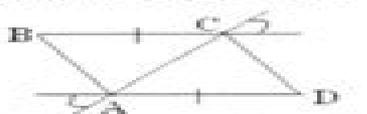
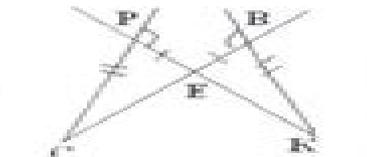
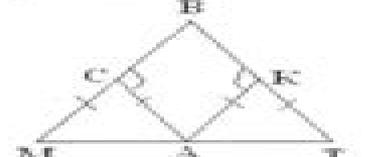
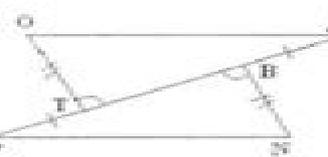
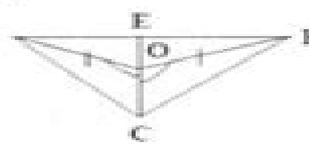


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Xiaolong Qi



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web 1 place 30 ml of your 0.60 M acetic acid in a clean 100 ml beaker measure the pH of the solution and record it in data table b as solution 1b table b pH data for acetate

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web explanation the equivalence point is the point during a titration when there are equal equivalents of acid and base in the solution since a strong acid will have more effect on