

9. (a) Calculate the surface tension of benzene. Given : Density of benzene at  $20^{\circ}\text{C}$  is  $0.878\text{g/ml}$ . The parachor of C, H, double bond are 4.8, 17.1, 23.2 respectively. 3, 3, 2
- (b) Calculate the parachor for hexane. Given : Parachor of ethane, propane are 110.5 and 150.8 respectively.
- (c) Find parachor values of H in decane ( $\text{C}_{10}\text{H}_{22}$ ). Given :  $[P]_{\text{CH}_2} = 39.0$ ;  $[P]_{\text{C}_{10}\text{H}_{22}} = 424.2$ .

# Gradechemistry November 2014 Paper

**ML Morrison**



**Gradechemistry November 2014 Paper:**

## The Enigmatic Realm of **Gradechemistry November 2014 Paper**: Unleashing the Language is Inner Magic

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