

1. In which one of the following regions was Dhanyakataka, which flourished as a prominent Buddhist centre under the Mahasanghikas, located?

- (a) Andhra
- (b) Gandhara
- (c) Kalinga
- (d) Magadha

2. With reference to ancient India, consider the following statements :

1. The concept of Stupa is Buddhist in origin.
2. Stupa was generally a repository of relics.
3. Stupa was a votive and commemorative structure in Buddhist tradition.

How many of the statements given above are correct?

- (a) Only one
- (b) Only two
- (c) All three
- (d) None

3. With reference to ancient South India, Korkai, Poempuhar and Muchiri were well known as

- (a) capital cities
- (b) ports
- (c) centres of iron-and-steel making
- (d) shrines of Jain Tirthankaras

4. Which one of the following explains the practice of 'Vattakirutal' as mentioned in Sangam poems?

- (a) Kings employing women bodyguards
- (b) Learned persons assembling in royal courts to discuss religious and philosophical matters
- (c) Young girls keeping watch over agricultural fields and driving away birds and animals
- (d) A king defeated in a battle committing ritual suicide by starving himself to death

5. Consider the following dynasties :

1. Hoysala
2. Gahadavala
3. Kakatiya
4. Yadava

How many of the above dynasties established their kingdoms in early eighth century AD?

- (a) Only one
- (b) Only two
- (c) Only three
- (d) None

6. With reference to ancient Indian History, consider the following pairs :

Literary work	Author
1. Devichandragupta	: Bilhana
2. Hammira-Mahakavya	: Nayachandra Suri
3. Milinda-panha	: Nagarjuna
4. Nitivakyamrita	: Somadeva Suri

How many of the above pairs are correctly matched?

- (a) Only one
- (b) Only two
- (c) Only three
- (d) All four

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**Elisabeth Pereira, Carlos Costa, Zélia
Breda**



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web aug 15 2020 the equilibrium formulated in scheme 1 between hydrogen ion and the corresponding acid and base can be called a simple acid base equilibrium by mixing two simple systems a double acid base system and an acid base equilibrium result that can always be formulated as follows $\text{acid}_1 + \text{base}_2 \rightleftharpoons \text{acid}_2 + \text{base}_1$

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species that is capable of accepting a proton which requires a lone pair of electrons to bond to the H⁺ water is amphoteric which means it can act as both a Brønsted-Lowry acid and a Brønsted-Lowry base

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