## PAPER-T

## CHEMISTRY

### (SET BY NARAYANA EDUCATIONAL INSTITUTIONS LECTURERS)

1. The minimum angular momentum of an electron with the magnetic quantum numbers (-1, 0, +1)

(1)  $\frac{\sqrt{3}h}{\pi}$  (2)  $\frac{h}{\pi}$  (3)  $\frac{2}{h\pi}$  (4)  $\frac{3h}{\pi}$ 

Arrange the following molecules in the order of increasing bond angles H,S, NH<sub>a</sub>, CH<sub>a</sub>

NH<sub>s</sub> < H<sub>s</sub>S < CH<sub>s</sub>

(2) CH<sub>4</sub> < H<sub>2</sub>S < NH<sub>3</sub>

(3) H<sub>a</sub>S < NH<sub>a</sub> < CH<sub>a</sub>

, (4) NH<sub>3</sub> < CH<sub>4</sub> < H<sub>5</sub>S

Total kinetic energy of two moles of an Ideal Gas at 127°C is

(1) 8.31 kJ

(2) 9.98 kJ

(3) 3.20 kJ

(4) 5 68 kJ

Acetic acid undergoes diments alon in Benzene. Molecular weight pol accilion Previous 30.30.20/Estion Papes 26.97% acid in Benzene was observed as 90 in an experiment. Extent of association of acetic acid in the given solvent at the given condition is

(1) 66% (2) 33% (3) 90% (4) 100%

Ammonia can be dried over

(1) Quick lime (3) P.O.

(2) Slaked lime (4) Conc. H<sub>3</sub>SO<sub>4</sub>

CH, CN + H,O → X + NH, The density of 2M solution of X in water is 1.017 gm/ml. The molality of the solution is

(2) 1.55 (3) 2.26 (4) 1.25

Electrode potentials of various metals are given  $Fe \rightarrow Fe^{+2} + 2e^{-}$ ;  $E^{0} = + 0.44V$ 

 $Pb \rightarrow Pb^{+2} + 2e^{i0}$ ;  $E^{0} = + 0.13V$  $Ag^+ + e^{\otimes} \rightarrow Ag : E^0 = + 0.8V$ 

Cu+2 + 2e<sup>50</sup> → Cu: E<sup>0</sup> = + 0.34V

Which of the following oxidise Iron?

(1) Ag only

(2) Ag, Cu only

(3) Ag, Cu, Pb (4) Pb only

Ethyl alcohol is treated with chlorine. 8. The hybridisation of carbon atoms in final product is

(1) SP3, SP3

(2) SP3 SP2

(3) SP3, SP (4) SP2

9. α-chlorination of Acetic acid in presence of phosphorous is known as

(1) Schmidt reaction

(2) Hell-vohlard-Zelensky reaction

(3) Rosenmund Reaction

(4) Sandmeyer reaction

10. Fe+2 is attached to how many nitrogen in haemoglobin

(2) 5 (3) 6

The percentage of carbon in CO2 is

12. Formula of Deadburnt Gypsum is

(1) CaSO, 2H,O (2) CaSO,

(3) 2CaSO, H,O (4) 2CaSO, 3H,O

13. In Castner Kellner Process, the Cathode in the middle compartment

(1) Hg

(2) Graphite

(3) Iron

(4) Platinum

The O-O bond is not present in

(1) H<sub>o</sub>SO<sub>c</sub> (2) CrO<sub>c</sub>

(3) HCIO<sub>4</sub> (4) HNO<sub>4</sub>

 A mixture of CH<sub>4</sub> and NH<sub>3</sub> heated to 1000°C in the presence of Al,O, can gg il was

(1) CH<sub>3</sub>NH<sub>2</sub> (2) HCN

(3) (CH<sub>2</sub>)<sub>2</sub>NH (4) CH<sub>2</sub>CN

- In carius mothod halogens are estimated as (X = halogen)

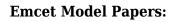
(1) X<sub>2</sub>

(2) BaX.



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**Scott C. Dulebohn** 



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