

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

## SOLUBILITY WORKSHEET

Use the solubility curves on the attached page to answer the following questions.

- These solubility curves are based on the solubility (in grams) of various salts dissolving in \_\_\_\_\_ grams of water. In general for solids, as temperature increases, the solubility of the salt \_\_\_\_\_.
- Most and least soluble at different temperatures:
  - Which salt is **least** soluble at 20°C? \_\_\_\_\_
  - Which salt is **most** soluble at 10°C? \_\_\_\_\_
  - For which salt does the solubility change the least? \_\_\_\_\_
- How many grams of each solute below is needed to saturate 100g of water at 80 °C to prepare the following solutions?
  - KNO<sub>3</sub> \_\_\_\_\_
  - KCl \_\_\_\_\_
  - NaCl \_\_\_\_\_
- How many grams of each solute below are needed to make a **saturated** solution at 50 °C?
  - KClO<sub>3</sub> \_\_\_\_\_
  - NaNO<sub>3</sub> \_\_\_\_\_
  - KNO<sub>3</sub> \_\_\_\_\_
- At what temperature are the following solutes equally soluble in 100g of water?
  - NaNO<sub>3</sub> and KNO<sub>3</sub> \_\_\_\_\_
  - KNO<sub>3</sub> and NaCl \_\_\_\_\_
  - NaCl and KCl \_\_\_\_\_
- The quantity of solute per 100 g of water used in making some solutions is listed below. Is the solution **saturated, unsaturated, or supersaturated**?

Solute	Temperature	
a. 103 g KNO <sub>3</sub>	60 °C	_____
b. 40 g KCl	40 °C	_____

# Mrkanessciencepage Solubility Ans Packet Answer Key

**J Rink**



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